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54 **6,7-Dihydro-7,7-disubstituted-khellins.**

57 Novel analogues of khellin are 4-methoxy, 9-methoxy or 4,9-dimethoxy-6,7-dihydro-7,7-disubstituted-5H-furo[3,2-g][1]benzopyran-5-ones. They can be useful as anti-atherosclerotic agents.

6,7-DIHYDRO-7,7-DISUBSTITUTED-KHELLINS

The present invention relates to novel analogs of a known pharmacological agent, khellin, also known as "visamin", and structurally related antiatherogenic furochromones and other benzopyrans. Chemically, khellin is a furochromone. Furochromones are characterized generally by the structural formula IV. Specifically, khellin is the furochromone of formula V, and is trivially named 7-methyl-4,9-dimethoxyfurochromone. Khellin and related furochromones are naturally-occurring substances and have been used in crude form as pharmacological agents for centuries. Khellin is an extract from the plant *Ammi visnaga*. This plant grows wild in Eastern Mediterranean countries. Aside from khellin, *Ammi visnaga* is also a source of at least three other known and characterized furochromones, specifically visnagin, khellinin, and ammiol.

As indicated above, khellin exhibits a wide variety of pharmacological actions, rendering this compound a useful agent for numerous pharmacological purposes. For a comprehensive, but somewhat dated, review of the chemistry and physiological action of khellin-related products, see the reports of Hutterer, C.P., et al., *Chem. Revs.* 48:543-79 (1951) and Aubertin, E., *J. Med. Bordeaux* 127:821-823 (1950).

One principal action of khellin is its ability to induce relaxation of smooth muscle tissues. Particularly, khellin is known as a potent dilator of coronary blood vessels. This potent coronary vasodilator activity of khellin renders the compound useful in the treatment of angina pectoris and other diseases characterized by coronary artery insufficiency. For a description of the use of khellin in the treatment of such diseases, see Osher, H.L., et al., "Khellin in the Treatment of Angina Pectoris", *The New England Journal of Medicine* 244:315 (1951). Also the effects of enteric-coated khellin on coronary artery insufficiency is reported by Best, M.M., et al., *J. Med. Sci.* 222:35-9 (1951). The ability of khellin to relax smooth muscle also extends to gastrointestinal smooth muscle where khellin has been demonstrated to inhibit peristalsis, thus indicating antidiarrhetic potential. See Raymond-Hamet, M., *Compt. Rend.*

238:1624-6 (1954). Khellin may also be useful for the treatment of gastrointestinal disorders exhibiting a spasmodic component, as suggested by Anrep, G.V., et al., *Amer. Heart J.* 37:531-542 (1949). Further the antispasmodic effects of khellin on the urethra is reported by Colombo, G., et al., *Arch. Sci. Med.* 97:71 (1954) and Montorsi, W., et al., *Presse Med.* 63:81 (1955).

The antispasmodic action of khellin also extends to bronchial smooth muscle, rendering khellin useful in the treatment of asthma and other hypoxic pulmonary diseases. In this regard, see Silber, E.N., et al., "The Effect of Khellin on Cardio-Pulmonary Function in Chronic Pulmonary Disease", published in 1951; Anrep, G.V., et al., "Therapeutic Uses of Khellin", *The Lancet*, April 26, 1947, pages 557-8.

Khellin has also been reported to exert a hypotensive effect in humans by Jordan, H., *Arzneimittel-Forsch* 8:141-3 (1958), and 7:82-5 (1957). An additional account of the hypotensive effect of khellin is provided by Lian, C., et al., *Acta. Cardiol. (Brussels)* 5:373-88 (1950). With respect to overall cardiac effects, however, khellin has been reported to exert a cardiac depressive activity. In this regard see Samaan, K., et al., *J. Roy. Egypt Med. Assoc.* 33:953 (1950) and *J. Pharm. Pharmacol.* 1:538-44 (1949).

In addition to its effect on gastrointestinal smooth muscle reported above, khellin is also known as a gastric antisecretory and antiulcer agent. In this regard, the gastric antisecretory activity of khellin is reported by LaBarre, J., *Compt. Rend. Soc. Biol.* 150:1806-7 (1956) and 150:598-9 (1956).

Numerous other miscellaneous properties of khellin are also reported. For an account of its anthelmintic activity see Baytop, O.T., *Folia, Pharm. (Turkey)* 1:48-9 (1949). For an account of the CNS depressant activity of khellin see Chen, G., *Proc. Soc. Exptl. Biol. Med.* 78:305-7 (1951). For an account of the cytostatic activity of khellin see Apffel, C.A., *Deut. Med. Wochschr.* 80:414-16 (1955). Finally, the spermicidal action of khellin is reported by Swayne, V.R., et al., *Amer. J. Pharm.* 125:295-8 (1953).

Khellin and numerous chemically related furochromones (and derivatives thereof) are also useful in treatment and prevention of atherosclerosis by methods described in U.S. Patent 4,284,569.

Atherosclerosis in mammals is a disease characterized by the deposition of atherosclerotic plaque on arterial walls. While

atherosclerosis exhibits many varied forms and consequences, typical consequences of atherosclerotic diseases include angina pectoris, myocardial infarction, stroke and transient cerebral ischemic attacks. Other forms of atherosclerotic diseases include certain peripheral
5 vascular diseases and other ischemias (e.g., bowel and renal).

Medical science now recognizes that certain forms of atherosclerosis may be preventable or reversible. Agents capable of preventing or reversing atherosclerosis are characterized as exhibiting antiatherosclerotic activity. Since serum lipids have a
10 recognized association with atherogenesis, an important class of antiatherosclerotic agents are those with serum lipid-modifying effects. Serum lipids implicated in atherogenesis include serum cholesterol, serum triglycerides, and serum lipoproteins.

With respect to serum lipoproteins, at least three different
15 classes of these substances have been characterized: high density lipoproteins (HDL's), low density lipoproteins (LDL's), and very low density lipoproteins (VLDL's). HDL's are often referred to as alpha-lipoproteins, while LDL's and VLDL's are referred to as betalipoproteins. The enhancement to HDL levels (hyperalpha-lipoproteinemic activity) is postulated to have direct antiatherosclerotic effects.
20 See Eaton, R.P., J. Chron. Dis. 31:131-135 (1978). In contrast, agents which reduce serum LDL's and serum VLDL's (hypobetalipoproteinemic agents) are also associated with antiatherogenic effects. See Haust, M.D., "Reaction Patterns of Intimal Mesenchyme to Injury
25 and Repair in Atherosclerosis:", Adv. Exp. Med. Biol. 43:35-57 (1974), which postulates that serum LDL is a factor in atherosclerotic lesion formation.

Numerous animal models have been developed for assessing anti-atherogenic activity. Principal among these are models for assessing
30 hypobetalipoproteinemic activity in the rat, antiatherosclerotic activity in the Japanese quail, and lipoprotein modifying activity in the monkey. For a description of the operation of the hypobetalipoproteinemic rat model, refer to the known methods of Schurr, P.E., et al., "High Volume Screening Procedure for Hypobetalipoproteinemia
35 Activity in Rats", Adv. Exp. Med. Biol. 67: Atherosclerotic Drug Discovery, pp. 215-229, Plenum Press (1975). For a description of the Japanese quail model, see Day, C.E., et al., "Utility of a Selected Line (SEA) of the Japanese Quail (*Coturnix Coturnix japonica*) for the

Discovery of New Anti-Atherosclerosis Drugs", Laboratory Animal Science 27:817-821 (1977).

A suitable primate model for assessing antiatherosclerotic activity of chemical compounds is found in the cynomolgus monkey. In these animals base-line values for VLDL's, LDL's, and HDL's can be determined by controlling diet over a period of several weeks and sampling plasma daily. After establishing control values, the effects of drug treatment are assessed by administering by gavage with a pre-determined series of doses of test compounds for a similar period (e.g., two weeks).

The khellin, the khellin-related products of *Ammi visnaga*, and related furochromones (and derivatives) described in U.S. Patent 4,284,569 are all characterized by pronounced antiatherogenic activity, rendering these compounds useful in the treatment and prophylaxis of atherosclerosis, atherogenic hyperlipoproteinemia (i.e., hypobetalipoproteinemia) and atherogenic hypolipoproteinemia (i.e., hypoalphalipoproteinemia), and the untoward consequences thereof. These compounds exhibit this useful pharmacological activity in both mammalian and non-mammalian species, including humans.

The patients susceptible to the development of atherosclerotic diseases and the untoward consequences thereof are particularly those physically asymptomatic patients manifesting one or more risk factors known to predispose one to disease development. Such risk factors are high serum cholesterol and serum triglycerides, hypertension, obesity, diabetes, and genetic predisposition. Patients manifesting two or more risk factors are deemed to be especially susceptible to atherosclerotic diseases. These khellin-related materials all exhibit pronounced oral pharmacologic activity. Accordingly, in using these compounds for the treatment of atherosclerosis, an oral route of administration, either by conventional oral dosage forms or by mixture with food, represents the preferred method of their systemic administration. Alternatively, however, these compounds may be administered by other convenient routes of administration whereby systemic activity is obtained. These other routes of administration would, accordingly, include rectal, vaginal, subcutaneous, intravenous, and like routes.

In humans, the preferred route of administration is oral, in the form of capsules or tablets containing the drug.

The patient or animal being treated must be given periodic doses of the drug in amounts effective to reduce atherogenic serum lipoproteins (e.g., betalipoproteins) or selectively enhance levels of antiatherogenic serum lipoproteins (e.g., enhancing alphalipoprotein levels). Such effective dosages are readily determined by methods known in the art. For example, small daily doses of the drug (e.g., 50-100 mg) may be administered initially with higher succeeding doses until levels of atherogenic or antiatherogenic serum lipoproteins are favorably affected. By this regimen, a compound is administered initially at doses as low as about 50 mg per patient per day, with increasing doses up to about 200 mg per patient per day. In the event the antiatherogenic response in a patient being treated at a dose of 200 mg per day is insufficient, higher doses are also utilized to the extent tolerance permits further increases in dose.

While the preferred dosage regimen is with single daily dosing of patients, also preferred for obtaining more uniform serum levels of drug are multiple dosages per day (e.g., up to 4-6 times daily). Accordingly, when 4 daily doses of drug are to be administered, each such dose may be about 50 mg per patient per dose (200-300 mg per patient per dose), or higher depending on tolerance.

Similar doses are employed in non-human mammals, e.g., 1-5 mg/kg/day.

4,9-Dimethoxyfurochromones are known in the art. Such known compounds include 7-ethyl, 7-phenyl, 7-propyl, and 7-ethoxycarbonyl analogs described by Schonberg, A., et al., JACS 72:1611-17 (1950); 7- γ -pyridyl analogs, described by Schonberg, A., JACS 77:5439 (1955); 7-furanyl analogs, described by Musante, C., et al., Pharmaco. (Pavie) Ed. Sci. 15:81-94 (1960); 7-carboxyaldehyde analogs, described by Mustafa, A., et al., J. Org. Chem. 26:886 (1961). Also, 6-substituted 4,9-dimethoxyfurochromones are known. See, for example, the compounds described by Abu-Shady, H., UAR J. Pharm. Sci. 11:283 (1970).

4-Methoxy-7-aminomethylenefurochromones are also known in the art. See Abu-Shady, H., et al, J. Pharm. Belg. 33:397 (1978).

A wide variety of antiatherosclerotic furochromones are described in U.S. Patent 4,284,569.

PRIOR ART

Extensive pharmacological uses for khellin and related substances are known, as indicated above. Khellin analogs are also known in the art, as indicated above. See especially U.S. Patent 4,284,569 and the review by Mustafa, A., "Fuopyrans and Fuopyrones," John Wiley and Sons, Inc., N.Y., N.Y. (1967), pp. 102-159 (Chapter III: Furochromones). Also see U.S. Patent 2,680,119 describing 6- and/or 7-substituted furochromones, i.e., alkyl, alkoxyalkyl and phenylalkyl substituted compounds.

SUMMARY OF THE INVENTION

10 The present invention particularly provides:

(1) A dihydrofurochromone of formula I

wherein R_1 and R_2 , being the same or different, are individually:

- (a) C_1 - C_6 alkyl,
- (b) trifluoromethyl,
- 15 (c) C_5 - C_{10} cycloalkyl with the proviso that the cycloalkyl ring is C_5 - C_7 ,
- (d) C_2 - C_8 alkylaminoalkyl,
- (e) C_2 - C_8 alkoxyalkyl,
- (f) C_2 - C_8 alkylthioalkyl,
- 20 (g) C_2 - C_8 alkylsulfinylalkyl,
- (h) C_2 - C_8 alkylsulfonylalkyl,
- (i) C_7 - C_{12} phenoxyalkyl optionally substituted on the phenyl ring by one, 2, or 3,
 - (i) hydroxy,
 - 25 (ii) C_1 - C_3 alkoxy,
 - (iii) C_1 - C_3 alkyl,
 - (iv) trifluoromethyl,
 - (v) halo which is fluoro, chloro, or bromo, with the proviso that not more than two such substituents are other than alkyl,
 - 30 (j) C_7 - C_{12} phenylthioalkyl optionally substituted on the phenyl ring by one, 2, or 3,
 - (i) hydroxy,
 - (ii) C_1 - C_3 alkoxy,
 - (iii) C_1 - C_3 alkyl,
 - 35 (iv) trifluoromethyl,
 - (v) halo which is fluoro, chloro, or bromo, with the proviso that not more than two such substituents are other than alkyl,
 - (k) phenyl optionally substituted by one, 2, or 3,

- (i) hydroxy,
(ii) C₁-C₃ alkoxy,
(iii) C₁-C₃ alkyl,
(iv) trifluoromethyl,
5 (v) halo which is fluoro, chloro, or bromo, with the proviso that not more than two such substituents are other than alkyl,
(l) aralkyl optionally substituted on the aromatic ring by one, 2, or 3,
(i) hydroxy,
10 (ii) C₁-C₃ alkoxy,
(iii) C₁-C₃ alkyl,
(iv) trifluoromethyl,
(v) halo which is fluoro, chloro, or bromo, with the proviso that not more than two such substituents are other than alkyl,
15 (m) 2- or 3-furanyl optionally substituted by
(i) hydroxy,
(ii) C₁-C₃ alkoxy,
(iii) C₁-C₃ alkyl,
(iv) trifluoromethyl,
20 (v) halo which is fluoro, chloro, or bromo, with the proviso that not more than two such substituents are other than alkyl,
(n) 2- or 3-thenyl optionally substituted by
(i) hydroxy,
(ii) C₁-C₃ alkoxy,
25 (iii) C₁-C₃ alkyl,
(iv) trifluoromethyl,
(v) halo which is fluoro, chloro, or bromo, with the proviso that not more than two such substituents are other than alkyl,
or
30 (o) -CH₂NR₈R₉ wherein R₈ and R₉, being the same or different, are individually,
(i) hydrogen,
(ii) C₁-C₈ alkyl,
(iii) C₅-C₁₀ cycloalkyl,
35 (iv) C₇-C₁₂ aralkyl, or
(v) phenyl optionally substituted by one, 2, or 3
a) hydroxy,
b) C₁-C₃ alkoxy,

c) C₁-C₃ alkyl,

d) trifluoromethyl,

e) halo which is fluoro, chloro, or bromo, with

the proviso that not more than two such substituents are other than
 5 alkyl, or wherein R₈ and R₉ are taken together with the nitrogen to
 form a saturated or unsaturated heterocyclic amine ring consisting of
 from 2 to 7 carbon atoms, inclusive, and zero, one, or 2 additional
 hetero atoms or hetero atom groups, with the proviso that said
 heterocyclic amine ring contains 4 to 8 atoms in the ring, said
 10 additional hetero atoms or hetero atom groups being selected from the
 group consisting of oxygen, nitrogen, sulfur, -SO-, and -SO₂- said
 heterocyclic amine ring being optionally substituted by C₁-C₄ alkyl,
 C₂-C₈ alkylthiomethyl, or C₂-C₈ alkoxymethyl or C₁-C₄ hydroxymethyl or
 phenyl; or

15 wherein R₁ and R₂ are taken together and form a bivalent moiety
 which is:

(a) -CH₂-(CH₂)_a-CH₂- wherein the integer "a" is zero to 5;

(b) -CH₂-(CH₂)_b-X-(CH₂)_c-CH₂-

wherein the integer "b" is zero and the integer "c" is zero, one,
 20 2, or 3 or the integer "b" is one and the integer "c" is zero, one, or
 2, and wherein X is oxa (-O-), thia (-S-), or -N(R₁₀)- wherein R₁₀ is

(i) hydrogen,

(ii) C₁-C₈ alkyl,

(iii) C₅-C₁₀ cycloalkyl,

25 (iv) C₇-C₁₂ aralkyl, or

(v) phenyl optionally substituted by one, 2, or 3

a) hydroxy,

b) C₁-C₃ alkoxy,

c) C₁-C₃ alkyl,

30 d) trifluoromethyl,

e) halo which is fluoro, chloro, or bromo, with the

proviso that not more than two such substituents are other than alkyl;

wherein R₃ is hydrogen or C₁-C₄ alkoxy and R₄ is hydrogen or
 C₁-C₄ alkoxy, with the provisos that one of R₃ and R₄ is hydrogen only

35 when the other is other than hydrogen and R₃ and R₄ are the same or
 different.

(2) A dihydrofurochromone selected from the group consisting of:

(a) 6,7-Dihydro-4,9-dimethoxy-7,7-dimethyl-5H-furo[3,2-

- g][1]benzopyran-5-one,
 (b) 7-Ethyl-6,7-dihydro-4,9-dimethoxy-7-methyl-5H-furo[3,2-g][1]benzopyran-5-one,
 (c) 7,7-Diethyl-6,7-dihydro-4,9-dimethoxy-5H-furo[3,2-g][1]benzopyran-5-one,
 5 (d) 6,7-Dihydro-4,9-dimethoxy-7-methyl-7-phenyl-5H-furo[3,2-g][1]benzopyran-5-one,
 (e) 4',9'-Dimethoxy-spiro[cyclopentane-1,7'-[7H]furo[3,2-g][1]benzopyran]-5'(6'H)-one,
 10 (f) 4',9'-Dimethoxy-spiro[cyclohexane-1,7'-[7H]furo[3,2-g][1]benzopyran]-5'(6'H)-one,
 (g) 2',3',5',6'-Tetrahydro-4,9-dimethoxy-spiro[7H]furo[3,2-g][1]benzopyran-7,4'-[4H]thiopyran]-5(6H)-one,
 (h) 2',3',5',6'-Tetrahydro-4,9-dimethoxy-spiro[7H-furo[3,2-g][1]benzopyran-7,4'-[4H]thiopyran]-5(6H)-one, 1'-oxide,
 15 (i) 2',3',5',6'-Tetrahydro-4,9-dimethoxy-spiro[7H-furo[3,2-g][1]benzopyran-7,4'-[4H]thiopyran]-5(6H)-one, 1',1'-dioxide,
 (j) 6,7-Dihydro-4,9-dimethoxy-7-methyl-7-[(methylthio)-methyl]-5H-furo[3,2-g][1]benzopyran-5-one,
 20 (k) 6,7-Dihydro-4,9-dimethoxy-7-methyl-7-[(methylthio)-methyl]-5H-furo[3,2-g][1]benzopyran-5-one, oxide,
 (l) 6,7-Dihydro-4,9-dimethoxy-7-methyl-7-[(methylthio)-methyl]-5H-furo[3,2-g][1]benzopyran-5-one, dioxide, or
 (m) 4,9-Dimethoxy-1'-methyl-spiro[7H-furo[3,2-g][1]benzo-
 25 pyran-7,4'-piperidine]-5(6'H)-one.

The carbon atom content of various hydrocarbon containing moieties is indicated by a prefix designating the minimum and maximum number of carbon atoms in the moiety, i.e., the prefix C_i-C_j indicates a carbon atom content of the integer "i" to the integer "j" carbon atoms, inclusive. Thus, C_1-C_3 alkyl refers to alkyl of 1-3 carbon atoms, inclusive, or methyl, ethyl, propyl, and isopropyl.

With respect to the above, C_1-C_4 alkyl is methyl, ethyl, propyl, or butyl, including isomeric forms thereof. Similarly, C_1-C_6 alkyl is methyl, ethyl, propyl, butyl, pentyl, hexyl, and isomeric forms thereof.

Examples of cycloalkyl of 3 to 10 carbon atoms, inclusive, are cyclopropyl, cyclobutyl, cyclopentyl, cyclohexyl, cycloheptyl, cyclodecyl, cyclopropylmethyl, cyclobutylmethyl, cyclopentylmethyl,

cyclohexylmethyl, cycloheptylmethyl, 1- or 2-cyclopropylethyl, 1- or 2-cyclobutylethyl, 1- or 2-cyclopentylethyl, 1- or 2-cyclohexylethyl, 1-, 2-, 3- or 4-methylcyclohexyl, (bicyclo[3.1.1]hept-2-yl)methyl, and 6,6-dimethyl-(bicyclo[3.1.1]hept-2-yl)methyl.

- 5 Example of alkoxyaminoalkyl are methylaminomethyl, ethylamino-methyl, propylaminomethyl, butylaminomethyl, isopropylaminomethyl, isobutylaminomethyl, tert-butylaminomethyl, pentylaminomethyl, and n-hexylaminomethyl.

- 10 Examples of C₂-C₈ alkoxyalkyl are methoxymethyl, ethoxymethyl, propoxymethyl, butoxymethyl, pentoxymethyl, hexoxymethyl, and heptoxymethyl.

- 15 Examples of C₂-C₈ alkylthioalkyl are methylthiomethyl, ethylthio-methyl, propylthiomethyl, butylthiomethyl, isopropylthiomethyl, isobutylthiomethyl, tert-butylthiomethyl, pentylthiomethyl, and n-hexylthiomethyl.

 Examples of C₂-C₈ alkylsulfinylalkyl are methylsulfinylmethyl, ethylsulfinylmethyl, propylsulfinylmethyl, butylsulfinylmethyl, isopropylsulfinylmethyl, isobutylsulfinylmethyl, tert-butylsulfinylmethyl, pentylsulfinylmethyl, and n-hexylsulfinylmethyl.

- 20 Examples of C₂-C₈ alkylsulfonylalkyl are methylsulfonylmethyl, ethylsulfonylmethyl, propylsulfonylmethyl, butylsulfonylmethyl, isopropylsulfonylmethyl, isobutylsulfonylmethyl, tert-butylsulfonylmethyl, pentylsulfonylmethyl, and n-hexylsulfonylmethyl.

- 25 Example of C₆-C₁₂ phenoxyalkyl optionally substituted are phenoxy-methyl, phenoxyethyl, phenoxypropyl, phenoxybutyl, phenoxy-pentyl, 1-, 2-, or 3-methylphenoxy-methyl, 1-, 2-, or 3-hydroxyphenoxy-methyl, 1-, 2-, or 3-methoxyphenoxy-methyl, 1-, 2-, or 3-trifluoromethylphenoxy-methyl, 1-, 2-, or 3-fluorophenoxy-methyl, and 1-fluoro-3-methyl-phenoxy-methyl.

- 30 Example of C₆-C₁₂ phenylthioalkyl optionally substituted are phenylthiomethyl, phenylthioethyl, phenylthiopropyl, phenylthiobutyl, phenylthiopentyl, 1-, 2-, or 3-methylphenylthiomethyl, 1-, 2-, or 3-hydroxyphenylthiomethyl, 1-, 2-, or 3-methoxyphenylthiomethyl, 1-, 2-, or 3-trifluoromethylphenylthiomethyl, 1-, 2-, or 3-fluorophenylthiomethyl, and 1-fluoro-3-methyl-phenylthiomethyl.

35 Example of C₆-C₁₂ phenyl optionally substituted are phenyl, 1-, 2-, or 3-methylphenyl, 1-, 2-, or 3-hydroxyphenyl, 1-, 2-, or 3-methoxyphenyl, 1-, 2-, or 3-trifluoromethylphenyl, 1-, 2-, or 3-

fluorophenyl, and 1-fluoro-3-methylphenyl.

Examples of aralkyl of 7 to 12 carbon atoms are naphthylmethyl and haphthylethyl.

Examples of heterocyclic amines corresponding to heterocyclic
5 amine rings are thiazolidine, 3-piperidine methanol, 2-piperidine
methanol, piperidinic acid, 3-piperidine ethanol, 2-piperidine
ethanol, 1-piperizinepropanol, p-piperazinoacetoxypheone, 4-phenyl-
1,2,3,6-tetrahydropyridine, 4-phenylpiperidine, proline, 3-pyrrolidi-
10 lidine-4-carboxylic acid, thiomorpholine, nipecstamide, morpholine, 2-
methylpiperidine, 3-methylpiperidine, 4-methylpiperidine, N-methyl-
piperazine, and 1-methylhomopiperazine.

The compounds in accordance with the present invention are all
useful as antiatherosclerotic agents. Thus these compounds are
15 employed by methods known in the art for the use of khellin and
related furochromones in the treatment and prevention of athero-
sclerosis. Accordingly, compounds of formula I are employed in humans
and in nonhuman mammals at doses from about 0.1-50 mg/kg/day orally.
These compounds are used orally in conventional oral dosage forms,
20 including capsules, tablets, and pharmaceutically acceptable liquids.
Other routes of administration may also be employed, utilizing
equivalent dosages and the appropriate conventional dosage form for
the route of administration selected. Such alternatively dosage forms
include rectal, vaginal, subcutaneous, intravenous, and like routes of
25 administration.

Compounds of formula I are also useful as food or feed additives
whereby the ingestion of food or feed by the mammal being treated
results in an effective oral dose of the compound.

Preferred among the compounds of formula I are those wherein at
30 least one of R_3 and R_4 is methoxy and more preferably both are
methoxy.

By virtue of the ring structure of the compounds of the present
invention, these substances are named as either 4-alkoxy-, 9-alkoxy-,
or 4,9-dialkoxy- (depending upon whether, respectively, R_3 is alkoxy
35 and R_4 is hydrogen, R_4 is alkoxy and R_3 is hydrogen, or R_3 and R_4 are
both alkoxy), 6,7-dihydro-7,7-disubstituted-5H-furo[3,2-g][1]benzo-
pyran-5-ones when R_1 and R_2 are not taken together to form a spiro
ring. The compounds so described by formula I are 6,7-dihydro-spiro-

[1,7'-[7H]furo[3,2-g][1]benzopyran]-5'(6'H)-ones.

The novel compounds disclosed in the present specification are all prepared by methods described in Chart A.

With respect to the chart, the various substituents R₁, R₂, R₃ and R₄ are as defined above in the specification.

With respect to Chart A, the formula XXII compound is known. The preparation of this compound wherein R₃ or R₄ is methoxy or R₃ and R₄ are both methoxy is described in U.S. Patent 4,284,569, where its synthesis from khellin or related compounds is described. Preferably the formula XXI compound is prepared by the methods described in Appendix A or B, or EPA 83302524.0 or (also filed 5.5.83 \equiv USSN 378686).

In accordance with the procedure of Chart A, the formula XXI compound is reacted with a ketone of formula XXII wherein R₁ and R₂ are as defined above. This reaction produces the formula XXIII product, i.e., the compounds of formula I. This reaction proceeds by combining the ketone of formula XXII and the formula XXI reactant with a cyclic amine base, e.g., especially pyrrolidine, and heating for a prolonged period at elevated temperature. For example, the reaction proceeds at reflux for a period ranging from several hours to several days. Thereupon the formula XXIII product is obtained from the reaction mixture by conventional separation techniques, i.e., filtration, chromatography, and crystallization.

The sulfoxide (-SO-) and sulfone (-SO₂-) compounds herein are preferably prepared from corresponding thio (-S-) compounds by oxidation. Such oxidizations are carried out by methods known in the art for the preparation of sulfoxides and sulfonyl derivatives of thio-containing compounds. For example, a useful oxidizing agent for this purpose is m-chloroperoxybenzoic acid and sodium periodate.

Thus, in accordance with the procedure of Chart A, each of the various formula I compounds provided in the present disclosure is prepared.

DESCRIPTION OF THE PREFERRED EMBODIMENTS

The preparation of the novel compounds disclosed herein is more readily understood by the operation of the following examples:

Example 1: 6,7-Dihydro-4,9-dimethoxy-7,7-dimethyl-5H-furo[3,2-g][1]benzopyran-5-one (Formula I, R₁ and R₂ are both methyl, and R₃ and R₄ are both methoxy).

Refer to Chart A.

Toluene (15 ml) and 1-(6-hydroxy-4,7-dimethoxy-5-benzofuranyl)-ethanone (6.5 g) are combined with acetone (25 ml) and pyrrolidine (0.75 ml). The resulting mixture is then heated under nitrogen to reflux for 96 hr and then allowed to cool to ambient temperature. 5 Concentration under reduced pressure yields a dark oil which crystallizes. High pressure liquid chromatography on 350 g of silica gel eluting with ethyl acetate yields 8.4 g of title product as a yellow solid. This solid is diluted with dichloromethane and filtered through basic alumina. The filtrate is then dried and the resulting 10 residue recrystallized from ethyl acetate in hexane to yield 3.8 g of pure title product, melting point 90-92°C. Silica gel TLC R_f is 0.21 in 25% ethyl acetate in hexane. IR absorptions (cm^{-1}) are observed at 3140, 1670, 1610, 1595, 1550, 1480, 1355, 1305, 1145, 1080, and 1060. NMR absorptions (CDCl_3 , δ) are observed at 7.55, 6.95, 4.05, 4.01, 15 2.72, and 1.50.

Example 2: 7-Ethyl-6,7-dihydro-4,9-dimethoxy-7-methyl-5H-furo[3,2-g][1]benzopyran-5-one (Formula I, R_1 is methyl, R_2 is ethyl, and R_3 and R_4 are both methoxy).

Refer to Chart A.

20 Benzene (100 ml) and 1-(6-hydroxy-4,7-dimethoxy-5-benzofuranyl)-ethanone (9.68 g) and 2-butanone (10 ml) are combined under a nitrogen atmosphere and pyrrolidine (1 ml) is added. The resulting mixture is then heated to reflux for 72 hr. After cooling to ambient temperature, the reaction mixture is diluted with ethyl acetate and extracted 25 with 2N aqueous sodium hydroxide (50 ml), 2N hydrochloric acid (100 ml), and brine (20 ml). The organic phase is then dried over magnesium sulphate and filtered. The resulting filtrate is then concentrated under reduced pressure to a residue which is chromatographed on 330 g of silica gel, eluting with 20% ethyl acetate in hexane. Pure 30 title product, 8.0 g of an oil, is obtained. Crystallization of the oil followed by recrystallization yields 4.04 g of pure title product, melting point 68-71°C. Silica gel TLC R_f is .22 in 25% ethyl acetate in hexane. IR absorptions (cm^{-1}) are observed at 3180, 3140, 1685, 1615, 1590, 1550, 1350, 1300, 1140, 1080, and 1060. NMR absorptions 35 (CDCl_3 , δ) are observed at 7.50, 6.90, 4.04, 4.01, 2.73, 1.82, 1.45, and 1.00.

Example 3: 7,7-Diethyl-6,7-dihydro-4,9-dimethoxy-5H-furo[3,2-g][1]benzopyran-5-one (Formula I, R_1 and R_2 are both

ethyl, and R₃ and R₄ are both methoxy).

Refer to Chart A.

Following the procedure of Example 1 or Example 2, 1-(6-hydroxy-4,7-dimethoxy-5-benzofuranyl)-ethanone (5.1 g) and 3-pentanone (12.8 ml) are transformed to the title product, 2.82 g of crystalline material, melting point 72-73°C. Silica gel TLC R_f is .24 in 25% ethyl acetate in hexane. IR absorptions (cm⁻¹) are observed at 3120, 3060, 1670, 1605, 1595, 1500, 1480, 1345, 1315, 1140, 1080, 1065, and 750. NMR absorptions (CDCl₃, δ) are observed at 7.50, 6.92, 4.07, 2.75, 1.81, and 0.98.

Example 4: 6,7-Dihydro-4,9-dimethoxy-7-methyl-7-phenyl-5H-furo[3,2-g][1]benzopyran-5-one (Formula I, R₁ is methyl, R₂ is phenyl, and R₃ and R₄ are both methoxy).

Refer to Chart A.

Following the procedure of Example 1 or Example 2, 1-(6-hydroxy-4,7-dimethoxy-5-benzofuranyl)-ethanone (5.3 g) and acetophenone (6.0 ml) are transformed to 0.5 g of crystalline title product, melting point 161-162°C. Silica gel TLC R_f is .10 in 25% ethyl acetate in hexane. IR absorptions (cm⁻¹) are observed at 3130, 1675, 1600, 1550, 1495, 1350, 1315, 1310, 1275, 1140, 1080, 1055, 770, 740, 730, and 705. NMR absorptions (CDCl₃, δ) are observed at 7.4, 6.84, 4.15, 3.95, 3.38, 3.04, and 1.80.

Example 5: 4',9'-Dimethoxy-spiro[cyclopentane-1,7'-[7H]furo[3,2-g][1]benzopyran]-5'(6'H)-one (Formula I, R₁ and R₂ taken together are -(CH₂)₄-, and R₃ and R₄ are both methoxy).

Refer to Chart A.

Following the procedure of Example 1 or Example 2, 1-(6-hydroxy-4,7-dimethoxy-5-benzofuranyl)-ethanone (50 g) and cyclopentanone (29 ml) are transformed to title product, 49.2 g of a yellow solid, melting point 75-77°C. Silica gel TLC R_f is 0.32 in 25% ethyl acetate in hexane. IR absorptions (cm⁻¹) are observed at 3120, 3070, 1685, 1610, 1590, 1550, 1480, 1355, 1145, and 1080. NMR absorptions (CDCl₃, δ) are observed at 7.55, 6.93, 4.08, 4.00, 2.83, 2.30-1.50.

Example 6: 4',9'-Dimethoxy-spiro[cyclohexane-1,7'-[7H]furo[3,2-g][1]benzopyran]-5'(6'H)-one (Formula I, R₁ and R₂ taken together are -(CH₂)₅-, and R₃ and R₄ are both methoxy).

Refer to Chart A.

Following the procedure of Example 1 or Example 2, 1-(6-hydroxy-4,7-dimethoxy-5-benzofuranyl)-ethanone (5.97 g) and cyclohexanone (3.45 ml) are transformed to pure title product as a pale yellow crystal (5.93 g), melting point 102-104°C. Silica gel TLC R_f is .30 in 25% ethyl acetate in hexane. IR absorptions (cm^{-1}) are observed at 3140, 3120, 1690, 1620, 1540, 1480, 1350, 1280, 1125, and 1060. NMR absorptions (CDCl_3 , δ) are observed at 7.52, 6.92, 4.08, 2.70, 2.25-1.20.

10 Example 7: 2',3',5',6'-Tetrahydro-4,9-dimethoxy-spiro[7H-furo[3,2-g][1]benzopyran-7,4'-[4H]thiopyran]-5(6H)-one (Formula I, R_1 and R_2 taken together are $-(\text{CH}_2)_2\text{-S-}(\text{CH}_2)_2\text{-}$, and R_3 and R_4 are both methoxy).

Refer to Chart A.

15 Following the procedure of Example 1 or Example 2, 1-(6-hydroxy-4,7-dimethoxy-5-benzofuranyl)-ethanone (10.63 g) and tetrahydrothiopyran-4-one (7.8 g) are transformed to pure title product (9.42 g), melting point 146.7-147.9°C. Silica gel TLC R_f is .1 in 25% ethyl acetate in hexane. IR absorptions (cm^{-1}) are observed at 3140, 3120, 20 1685, 1620, 1535, 1475, 1350, 1125, and 1065. NMR absorptions (CDCl_3 , δ) are observed at 7.55, 6.95, 4.10, 4.07, 3.45, 2.80, 2.72, 2.7-1.5.

Example 8: 2',3',5',6'-Tetrahydro-4,9-dimethoxy-spiro[7H-furo[3,2-g][1]benzopyran-7,4'-[4H]thiopyran]-5(6H)-one, 1'-oxide and 2',3',5',6'-Tetrahydro-4,9-dimethoxy-spiro[7H-furo[3,2-g][1]benzopyran-7,4'-[4H]thiopyran]-5(6H)-one, 1',1'-dioxide (Formula I: R_1 and R_2 taken together are $-(\text{CH}_2)_2\text{-SO-}(\text{CH}_2)_2\text{-}$ or $-(\text{CH}_2)_2\text{-SO}_2\text{-}(\text{CH}_2)_2\text{-}$, respectively, and R_3 and R_4 are both methoxy.

Refer to Chart A.

30 A. A solution of the title product of Example 7 (4.15 g) in a mixture of methanol and tetrahydrofuran (1:1) are diluted with water (75 ml) and then treated with sodium periodate (2.7 g). After about 30 min a precipitate appears. The reaction is then continued with stirring at ambient temperature for 12 hr and filtered. The filter cake is then washed with methanol (100 ml) and the total filtrate concentrated under reduced pressure to a residue, a mixture of title product.

B. The mixture of Part A is then combined with dichloromethane

and m-chloroperoxybenzoic acid (2.1 g) and stirred for 16 hrs. Treatment with 2 normal sodium hydroxide (75 ml) is followed by stirring for 5 min and separation of phases. The organic layer is then dried over sodium sulfate and concentrated under reduced pressure to a residue. Chromatography on silica gel eluting with ethyl acetate in trichloromethane (1:1) yields 4 g of sulfone. Recrystallization from acetone and hexane (5:1) yields 3.6 g of pure title product. Silica gel TLC R_f is 0.33 in ethyl acetate. IR absorptions (cm^{-1}) are observed at 3160, 3120, 3080, 1685, 1605, 1595, 1555, 1475, 1315, 1295, 1135, 1100, 1080, and 1070. NMR absorptions (CDCl_3 , δ) are observed at 7.61, 6.99, 4.18, 4.13, 3.8-2.9, 2.78, and 2.75-2.2.

Example 9: 6,7-Dihydro-4,9-dimethoxy-7-methyl-7-[(methylthio)-methyl]-5H-furo[3,2-g][1]benzopyran-5-one (Formula I, R_1 is methyl, R_2 is methylthiomethyl, and R_3 and R_4 are both methoxy).

Refer to Chart A.

Following the procedure of Example 1 or Example 2, 1-(6-hydroxy-4,7-dimethoxy-5-benzofuranyl)-ethanone (8.33 g) and thiomethyl acetone (5.5 ml) are transformed to 7.5 g of crystalline title product. Recrystallization from methyl acetate in hexane yields pure crystalline product, melting point 64-66°C. Silica gel TLC R_f is .40 in 50% ethyl acetate in hexane. IR absorptions (cm^{-1}) are observed at 3160, 3120, 3060, 1680, 1615, 1590, 1545, 1475, 1305, 1160, 1150, 1140, 1080, and 1065. NMR absorptions (CDCl_3 , δ) are observed at 7.55, 6.95, 4.05, 4.01, 3.16, 2.93, 2.60, 2.26, and 1.58.

Example 10: 6,7-Dihydro-4,9-dimethoxy-7-methyl-7-[(methylthio)-methyl]-5H-furo[3,2-g][1]benzopyran-5-one, oxide and 6,7-Dihydro-4,9-dimethoxy-7-methyl-7-[(methylthio)-methyl]-5H-furo[3,2-g][1]benzopyran-5-one, dioxide (Formula I: R_1 is methyl, R_2 is methylsulfinylmethyl or methylsulfonylmethyl, respectively, and R_3 and R_4 are both methoxy).

Refer to Chart A.

Following the procedure of Example 8, the product of Example 7 is converted to a corresponding sulfoxide or sulfone.

Example 11: 4,9-Dimethoxy-1'-methyl-spiro[7H-furo[3,2-g][1]benzopyran-7,4'-piperidine]-5(6H)-one (Formula I, R_1 and R_2 taken together are $-(\text{CH}_2)_2\text{-N}(\text{CH}_3)\text{-(CH}_2)_2\text{-}$, and R_3 and

R₄ are both methoxy).

Refer to Chart A.

Following the procedure of Example 1 or Example 2, 1-(6-hydroxy-4,7-dimethoxy-5-benzofuranyl)-ethanone (7.35 g) and N-methyl-4-piperidone (6 ml) are transformed to 6.72 g of title product as a tan crystal, melting point 102-103.1°C. Silica gel TLC R_f is .21 in 10% methanol in trichloromethane. IR absorptions (cm⁻¹) are observed at 3100, 3060, 2800, 2780, 2760, 1685, 1760, 1625, 1610, 1590, 1545, 1540, 1480, 1150, 1130, and 1075. NMR absorptions (CDCl₃, δ) are observed at 7.51, 6.94, 4.10, 4.07, 2.72, 2.7-2.43, 2.35, and 2.2-1.6.

In accordance with the procedures of the examples given above, there are prepared each of the various novel Formula I compounds disclosed herein.

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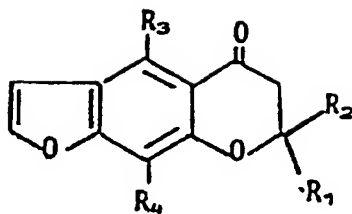
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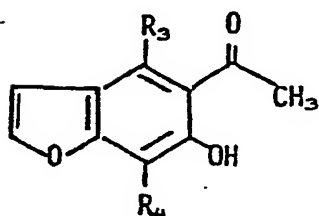
FORMULAS

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I

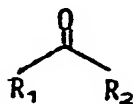
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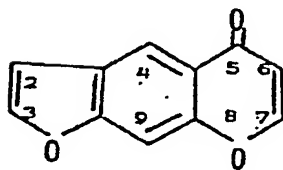
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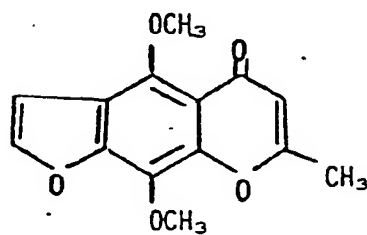
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IV

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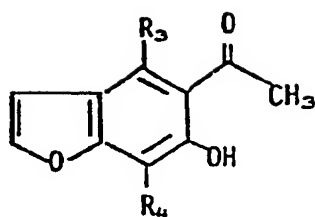


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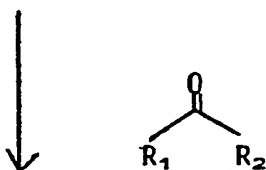
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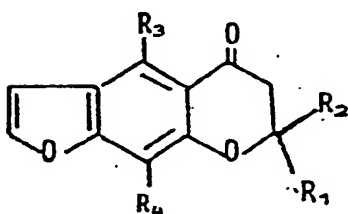
XXI

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XXII

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XXIII

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APPENDIX A
DESCRIPTION

BACKGROUND OF THE INVENTION

The present invention provides novel compositions of matter and processes for their preparation. Particularly, the present invention relates to novel chemical intermediates and associated processes for the preparation of furochromones. Most especially, the present invention provides for the preparation of novel antiatherosclerotic furochromones, particularly khellin analogs.

Khellin and related compounds are known to exert a wide variety of pharmacological effects. Recently, khellin has been reported to exhibit useful antiatherosclerotic activities. Moreover, numerous analogs of khellin likewise are known to exert useful antiatherosclerotic effects. For example, 7-methylthiomethyl-4,9-dimethoxy-furochromone is described in United States Patent 4,284,569 as such a useful antiatherosclerotic substance.

Methods for the total synthesis of khellin are known. For example, pyrogallol has been employed as a starting material for the synthesis of furochromones such as khellin. See Clarke, J.R., et al., J. Chem. Soc., 302 (1949), Baxter, R.A., et al., J. Chem. Soc., S30 (1949), Schonberg, A., et al., J. Am. Chem. Soc., 73:2960 (1951), Murti, V.V.S., et al., Proc. of the Indian Acad. of Sci., 30A:107 (1949), and Geissman, T.A., et al., J. Am. Chem. Soc., 73:1280 (1951). Also descriptive of the synthesis of khellin are Spath, E., et al., Chem. Ber., 71:106 (1938), Dann, O., et al., Chem. Ber., 93:2829 (1960), Dann, O., et al., Ann. Chem., 605:146 (1957), and Murti, V.V.S., et al., J. Sci. Ind. Res. (India), 8B:112 (1949). See also United States Patent 2,680,119 describing the synthesis of khellin and related compounds.

Other references describing the synthesis of intermediates useful in the preparation of khellin for analogs include: Aneja, R., et al., Chem. Ber., 93:297 (1960), Aneja, R., et al., J. Sci. Ind. Res. (India), 17B:382 (1958), Gardner, T.S., et al., J. Org. Chem., 15:841 (1950), and Rowe, L.R., et al., Indian J. Chem., 5:105 (1967).

Accordingly, the references cited above describe the preparation of 1-(6-hydroxy-4,7-dimethoxy-5-benzofuranyl)-ethanone. Also known is the related compound 6-hydroxy-4,7-dimethoxy-5-benzofurancarboxylic acid, methyl ester, described by Musante, C., Gazz. Chim. Ital.,

88:910 (1958).

PRIOR ART

Methods of the total synthesis of khellin are known, as are certain chemical intermediates useful in its synthesis.

- 5 Most typically, however, the total synthesis of furochromones from benzofurans has been accomplished by utilizing a substituted benzene ring from which to synthesize the fused benzofuran ring system. See Mustafa, A., "Benzofurans," John Wiley and Sons, 1974, and Mustafa, A., "Furopyrans and Fuopyrones, Chapter 3:
10 Furochromones," John Wiley and Sons, New York, New York, 1967.

United States Patent 4,284,569 provides a variety of novel anti-atherosclerotic furochromones.

SUMMARY OF THE INVENTION

The present invention particularly provides:

- 15 (a) A process for preparing a compound of formula XI which comprises:

- (1) reacting the lithium dianion of a compound of formula X with succinic anhydride;
- (2) C₁-C₄ alkyl esterfying the resulting formula XII
20 ketodiacid of step (1);
- (3) reacting the resulting formula XIII ketodiester of step (2), wherein R₁₁ is C₁-C₄ alkyl, with an amide acetal of formula XIV, wherein R₃ and R₄, being the same or different, are C₁-C₄ alkyl;
- (4) cyclizing of the resulting formula XV compound of step
25 (3), wherein R₃, R₄ and R₁₁, are as defined above;
- (5) dialkylating the resulting formula XVI benzofuran of step (4), wherein R₁₁ is as defined above;
- (6) oxidizing the resulting formula XVII dialkoxybenzofuran of step (5), wherein R₁ is C₁-C₄ alkyl and R₁₁ is as defined above;
30 and
- (7) reducing the resulting formula XVIII compound of step (6), wherein R₂ and R₁₁ are as defined above, to the formula XI compound;

- 35 (b) A furochromone intermediate of formula I or II, wherein R₁₁ is C₁-C₄ alkyl; wherein R₂ is hydrogen or C₁-C₄ alkyl; and wherein R₁₁ is as defined above; wherein W is α -H: β -H or =CH-NR₃R₄; wherein R₃ and R₄, being the same or different, are as defined above;

- (c) A furochromone intermediate of formula III;

(d) A furochromone intermediate of formula IV, wherein R₁₁ is as defined above;

(e) A furochromone intermediate of formula V, wherein R₃, R₄ and R₁₁ are as defined as above;

5 (f) A furochromone intermediate of formula VI, wherein R₁₁ is as defined above;

(g) A furochromone intermediate of formula VII, wherein R₁₁ is as defined above;

10 (h) A furochromone intermediate of formula I or II, wherein R₁, R₃, R₄, and R₁₁ are all methyl; and

(i) An anti-atherosclerotic furochromone of formula VIII wherein R₁₀ is C₂-C₄ alkyl; wherein R₁₂ is:

- 15 (1) hydrogen;
(2) C₁-C₈ alkyl;
(3) C₂-C₈ alkoxymethyl;
(4) C₂-C₈ alkylthioalkyl;
(5) trifluoromethyl;
(6) phenoxymethyl optionally substituted by chloro, fluoro,
20 trifluoromethyl, C₁-C₃ alkyl or C₁-C₃ alkoxy;

(7) phenylthiomethyl optionally substituted by chloro, fluoro, trifluoromethyl, C₁-C₃ alkyl or C₁-C₃ alkoxy;

(8) -CH₂-S(O)_n-R₂₀, wherein n is zero, one, or 2 and R₂₀ is C₁-C₅ alkyl; or

25 (9) -CH₂NR₈R₉, wherein R₈ and R₉ are hydrogen, C₁-C₁₂ alkyl or wherein R₈ and R₉, taken together with N, form a saturated or unsaturated heterocyclic amine ring consisting of from 2 to 7 carbon atoms, inclusive, and zero, one, or 2 additional hetero atoms, with the proviso that said heterocyclic amine ring contains 4 to 8 atoms in
30 the ring, said additional hetero atoms being selected from the group consisting of oxygen, nitrogen, and sulfur, said heterocyclic amine ring being optionally substituted by C₁-C₄ alkyl, C₂-C₈ alkylthio-methyl or alkoxymethyl C₁-C₄ hydroxyalkyl, or phenyl;

wherein R₁₃ is:

- 35 (1) hydrogen;
(2) chloro, iodo, or bromo; or
(3) -CH₂-S(O)_n-R₂₀ wherein n and R₂₀ are as defined above, with the proviso that R₁₃ is -CH₂-S(O)_n-R₂₀ only when R₁₄ is methyl.

In accordance with the method described above, there is prepared the formula XI benzofuran when R₁ is methoxy. This formula XI benzofuran is known to be useful in the preparation of a wide variety of antiatherosclerotic substances, including khellin and various analogs thereof. See United States Patent 4,284,569. Similarly there are prepared the novel formula XI benzofurans when R₁ is C₂-C₄ alkoxy. These intermediates are useful in the preparation of novel anti-atherosclerotic 4,9-di-(C₂-C₄)-alkoxy-furochromones of formula VIII by means described in United States Patent 4,284,569 for the preparation of the corresponding 4,9-dimethoxyfurochromones therein. Moreover, the manner of use of the novel 4,9-di-(C₂-C₄)-alkoxy-furochromones of formula VIII in the treatment and prevention of atherosclerosis is the same as that described in United States Patent 4,284,569 for the corresponding 4,9-dimethoxy compounds. Accordingly, the manner of the preparation and pharmacological use of these novel formula VIII compounds is incorporated herein by and reference from the description of the preparation and use in United States Patent 4,284,569 of the anti-atherosclerotic 4,9-dimethoxyfurochromones. Among the novel formula VIII compounds herein, the 4,9-diethoxyfurochromones are preferred.

The process of the present invention is more completely understood by reference to the charts below. In these charts, R₁, R₃, R₄, R₁₁, R₁₂, and R₁₃ are as defined above. R₅ is:

- (a) hydrogen;
- (b) C₁-C₈ alkyl;
- (c) C₂-C₈ alkoxymethyl;
- (d) C₂-C₈ alkylthioalkyl;
- (e) trifluoromethyl;
- (f) phenoxymethyl;
- (g) phenylthiomethyl;
- (h) phenoxymethyl or phenylthiomethyl, either of which is optionally substituted by one chloro, fluoro, trifluoromethyl, C₁-C₃-alkyl, or C₁-C₃-alkoxy; or
- (i) C₃-C₁₀ cycloalkyl.

With regard to the substituent W in formula I, this moiety is defined as either =CH-NR₃R₄ or α-H:β-H. In the latter case reflects the fact that each of the valences of the moiety W is a hydrogen atom, one of which is attached in the α configuration with respect to the

ring and the other which is attached in the β configuration with respect to the ring.

The carbon atom content of various hydrocarbon containing moieties is indicated by a prefix designating the minimum and maximum number of carbon atoms in the moiety, i.e., the prefix C_i-C_j indicates a carbon atom content of the integer "i" to the integer "j" carbon atoms, inclusive. Thus, C_1-C_3 alkyl refers to alkyl of 1-3 carbon atoms, inclusive, or methyl, ethyl, propyl, and isopropyl.

With respect to the charts, Chart A provides a method whereby the known formula XXI compound, 3-furoic acid is transformed to the highly functionalized benzofuran intermediate of formula XXVI useful in the synthesis of khellin and khellin analogs.

With further respect to chart A, the formula XXII compound is prepared from the formula XXI compound by first preparing the dianion of the formula XXII compound. See Knight, D.W., et al., J. Chem. Soc. Perkins, 1125 (1981), and references cited therein. Accordingly, this dianion is generated by treatment of the formula XXII compound with two equivalents of lithium diisopropylamide at low temperature, e.g., -78°C . At this temperature the resulting dianion is stable for several hours.

The dianion is then transformed to the formula XXII compound by treatment with succinic anhydride.

The resulting formula XXII is then esterified to the formula XXIII compound by conventional means. For example, an ethereal diazoalkane is employed or, for larger scale synthesis, an alkanol in hydrochloric acid is useful.

The formula XXIII compound is then converted to the formula XXIV compound by reaction at elevated temperature with an N,N-dialkylformamide dimethylacetal. Preferably, N,N-dimethylformamide dimethylacetal is employed at temperatures in excess of 80°C . For example, reaction at 100°C for 2 hr yields the formula XXIV compound.

Although this reaction preceeds in relatively high yield at elevated temperature, a less complex mixture of products is obtained by reacting the formula XXIII with the desired formamide for prolonged periods, i.e., up to a week. The optimal conditions for the preparation of the formula XXIV compound by this method are the stirring of the formula XXIII reactant in neat N,N-dimethylformamide dimethylacetal employing a trace of p-toluenesulfonic acid. Alternatively

base catalysis is employed using alkoxides, e.g., potassium tert-butoxide in an organic solvent.

The formula XXV compound is then prepared from the formula XXIV compound by a Dieckmann cyclization. Preferably the cyclization
5 occurs under basic conditions, e.g., preferably using potassium tert-butoxide in organic solvent, followed by an acid quench. Suitable organic solvents include benzene and tert-butanol, although the preferred solvent is tetrahydrofuran.

Finally, the formula XXVI compound of Chart A is prepared from
10 the formula XXV compound by alkylation. Alkylation can be accomplished quantitatively by treatment of the formula XXV reactant with an alkyl iodide in potassium carbonate.

Chart B provides a method whereby the formula XXXI compound, prepared as the formula XXVI compound of Chart A is converted to
15 formula XXXIII intermediate, a compound known to be useful in the preparation of both khellin and analogs thereof. In accordance with the procedure of Chart B, the formula XXXI compound is transformed to the formula XXXII compound by a Baeyer Villager oxidation. For this oxidation, m-chloroperbenzoic acid is employed at ambient temperature
20 in an organic solvent. Tetrahydrofuran or isopropanol represents the preferred solvent for undertaking this oxidation.

Finally, the formula XXXII compound is transformed into the formula XXXIII compound by conversion of the ester group to a methyl ketone. For this purpose a Grignard reagent in the presence of a
25 tertiary amine is employed according to the method of Kikkawa, I., and Yorifuji, T., Synthesis, 887 (1981).

Chart C provides a summary of the method by which the formula XLI compound, prepared as the formula XXXIII compound of Chart B, is transformed to khellin or analogs thereof. The procedures of Chart C
30 are, for example, known in the art from United States Patent 4,284,569, wherein Charts A-D of that patent describe the synthesis of the various formula XLII and formula XLIII compounds from the formula XLI starting material.

Accordingly, the charts herein provide a description of the
35 preparation and use of the novel process and compounds of the present invention.

DESCRIPTION OF THE PREFERRED EMBODIMENTS

The present disclosure is more fully understood by the operation

of the following examples:

Example 1: 3-Carboxy- γ -oxo-2-furanbutanoic acid (Formula XXII).

Refer to Chart A.

Diisopropylamine (202 g) is added to a flame dried 3 neck flask
5 with a mechanical stirrer, dropping funnel and nitrogen inlet. To this
amine is added tetrahydrofuran (THF, 600 ml). This solution is then
cooled to -78°C and *n*-butyllithium (128 g in hexane, 1.6 M) added over
20 min. After complete addition of the *n*-butyllithium the reaction is
stirred for 2.5 h, the last hour of which, the reaction vessel was
10 only 25% submerged in the dry ice bath. At this point, the lithium
diisopropylamine (LDH) separates from solution. Two liters of THF are
added and a homogeneous solution develops. The reaction is then
completely submerged in the dry ice bath and the 3-furoic acid,
formula XXI (100 g), in THF (600 ml) is added over 30 minutes. After
15 complete addition of 3-furoic acid the reaction is stirred an
additional two hours. At this point, succinic anhydride (100 g) in
THF (800 ml) is rapidly added via the addition funnel. A solid almost
immediately begins to separate from solution. As the reaction warms
to room temperature, quenching is effected with 2N hydrochloric acid
20 (3 liters). The entire reaction is poured into a portable separatory
funnel. The organic layer is separated and the aqueous back extracted
with trichloromethane (3 liters). The organic extracts are dried
(MgSO_4) and solvent removed in vacuo to give an off-white solid which
after an ether wash afforded 120 g of pure white title, mp $181-200^{\circ}\text{C}$.
25 IR absorption (cm^{-1}) absorptions are observed at 3140, 3120, 2740,
2640, 1740, 1705, 1630, 1615, 1570, 1330, 1270, 1215, 1165, 890 and
775. $^1\text{H-NMR}$ absorptions (δ , CDCl_3) are observed at 7.73, 7.05, 3.38
and 2.71. Mass spectral peaks are observed at 212, 194, 176, 167,
150, 149, 140, 139, 95, 55 and 39.

30 Example 2: 3-Carboxy- γ -oxo-2-furanbutanoic acid bis (methyl ester)
(Formula XXIII: R_{11} is methyl).

Refer to Chart A.

The formula XXII product of Example 1 (170 mg) is suspended in
trichloromethane (10 ml) and treated with excess diazomethane. When
35 TLC (5% EtOAc/ CHCl_3) indicated complete conversion to the bismethyl
ester, the trichloromethane is removed in vacuo to yield 211 mg of
crude product which was chromatographed (Merck B, 5% EtOAc/ CHCl_3) to
yield 211 mg of the pure title product as a colorless oil.

Silica Gel TLC R_f is 0.39 in 5% ethyl acetate in trichloromethane. IR absorptions (cm^{-1}) are observed at 3050, 1735, 1695, 1590, 1480, 1440, 1400, 1360, 1305, 1280 and 1160. $^1\text{H-NMR}$ absorptions (δ , CDCl_3) are observed at 7.5, 6.83, 3.90, 3.71, 3.30, and 2.75. Mass spectral peaks are observed at 240, 208, 181, 176, 153, 149, 123, 95, 55 and 38.

Example 3: β -[(Dimethylamino)methylene]-3-(methoxycarbonyl)- γ -oxo-2-furanbutanoic acid bis (methyl ester) (Formula XXIV: R_3 , R_4 and R_{11} are all methyl).

10 Refer to Chart A.

The formula XXIII product of Example 2 (500 mg) and N,N-dimethylformamide dimethylacetal (230 mg) are heated neat at 100°C for 1 hour. The reaction is then cooled to room temperature and excess acetal and methanol are removed in vacuo. The resulting brown oil is chromatographed (eluting with 5% methanol in ethyl acetate) to yield 280 mg of title product as a yellow oil.

Silica Gel TLC R_f is 0.4 in 5% methanol in ethyl acetate. IR absorptions (cm^{-1}) are observed at 3120, 2950, 1725, 1640, 1560, 1430, 1405, 1390, 1320 and 1160. NMR absorptions (δ , CDCl_3) are observed at 7.45, 6.95, 6.73, 3.82, 3.68 and 3.07. Mass spectral peaks are observed at 295, 263, 264, 237, 236, 218, 182, 153, 142 and 139.

Example 4: 6-Formyl-4,7-dihydroxy-5-benzofurancarboxylic acid methyl ester (Formula XXV: R_{11} is methyl).

Refer to Chart A.

25 A. (Preparation 1) Potassium metal (50 mg, 1.28 mmol) is added to tert-butanol (5 ml) under nitrogen with stirring. After the potassium metal has dissolved the formula (190 mg), in tert-butanol (5 ml) is added at room temperature. As the drops of diester hit the solution, a deep red color develops. This color slowly fades to yellow with time. After complete addition of the formula XXIV product of Example 3, the reaction is stirred an additional hour and then diluted with water. The reaction acidified with 2N HCl and was extracted with diethyl ether and then with trichloromethane. The combined organic extracts are dried (MgSO_4) and solvent removed in vacuo to yield a brown solid which after chromatography affords 40 mg of title product.

35 Silica Gel TLC R_f is 0.50 in 5% EtOAc/ CHCl_3 . IR absorptions (cm^{-1}) are observed at 3500, 2600, 1670, 1640, 1580, 1430, 1360, 1300, 1250. NMR absorptions (δ , CDCl_3) are observed at 10.5, 7.81, 7.0 and

4.0. Mass spectral peaks are observed at 236, 205, 204, 203, 176, 149, 148, 147, 119 and 63.

B. (Preparation 2) Potassium tert-butoxide (319 mg) is added to 20 ml of dry THF under nitrogen. This mixture is then cooled to -78°C and the formula XXIV starting material (420 mg) in THF (15 ml) added via syringe pump at a rate of 0.23 ml/min. A deep red color develops. After complete addition the reaction is stirred an additional 30 minutes and then the reaction is quenched at -78°C by the addition of 2N HCl. The reaction is then warmed to room temperature and poured into a separatory funnel. Then 2N HCl (50 ml) is added and the reaction was extracted with ethyl acetate (3 x 75 ml). The aqueous layer is then extracted with trichloromethane. The combined organic extracts are dried (MgSO₄) and solvent removed in vacuo to 340 mg of a brown solid. This solid is chromatographed over silica gel eluting with 5% ethyl acetate in trichloromethane to yield 200 mg of title product.

Example 5: 6-Formyl-4,7-dimethoxy-5-benzofurancarboxylic acid methyl ester (Formula XXVI: R₁ and R₁₁ are methyl).

Refer to Chart A.

6-Formyl-4,7-dihydroxy-5-benzofurancarboxylic acid methyl ester (Example 4, 4.70 g) is added to acetone (100 ml) followed by addition of methyl iodide (5.65 g) and potassium carbonate (5.0 g). The resulting mixture is heated at reflux for 24 hr. The reaction is cooled to room temperature and trichloromethane (100 ml) is added. Water (200 ml) is added. The organic layer is separated and the aqueous layer back extracted with trichloromethane (2 x 75 ml). The combined organic layer is dried (MgSO₄) and solvent removed in vacuo to yield a yellow oil. Chromatography over 100 g of silica gel eluting with 5% ethyl acetate in trichloromethane affords 5.2 g of title product as a pale yellow oil that slowly crystallized on standing. A pure product is prepared by recrystallization from methanol, mp 89.9-90.8°C.

Silica Gel TLC R_f is 0.44 in 5% EtOAc/CHCl₃. IR absorptions (cm⁻¹) are observed at 1730, 1680, 1600, 1470, 1440, 1390, 1340, 1305, 1290, 1060, 980 and 930. NMR absorptions (δ, CDCl₃) are observed at 10.4, 7.83, 6.97, 4.38 and 3.98. Mass spectral peaks are observed at 264, 249, 236, 233, 221, 205, 203, 189 and 147.

Example 6: 6-Hydroxy-4,7-dimethoxy-5-benzofurancarboxylic acid,

methyl ester (Formula XXXII: R₁ and R₁₁ are methyl).

Refer to Chart B.

5 A. (Procedure I) A solution of 6-Formyl-4,7-dimethoxy-5-benzofurancarboxylic acid, methyl ester (Example 5, 104.0 mg) in isopropanol (8.0 ml) is treated with 85% m-chloroperbenzoic acid (MCPBA, 188 mg) at ambient temperature and stirred overnight. The solvent is removed on the rotary evaporator and the residue is taken up into 10% aqueous sodium carbonate (10 ml) and diethyl ether (10 ml). After stirring for 30 minutes the layers are separated and the
10 aqueous is extracted with additional diethyl ether (1 x 20 ml). The ether extracts are combined and dried (MgSO₄). The residue is chromatographed (20 g silica gel) eluting with 20% ethyl acetate in isomeric hexanes (Skellysolve B) to provide 63.2 mg of white solid, title product, mp 82-4° (yield 63%).

15 B. (Procedure II) When the reaction is repeated as above substituting THF as the solvent, 101.5 mg of starting material and 184 mg of MCPBA yields 30.0 mg of pure product, mp 82-84°.

Example 7: 1-(6-Hydroxy-4,7-dimethoxy-5-benzofuranyl)-ethanone (Formula XXXIII: R₁ is methyl).

20 A. A 100 ml 3 neck flask is oven dried and cooled under nitrogen. Benzene (10 ml) is placed in the flask followed by methyl magnesium bromide (2.9 M in diethyl ether, 2.0 ml). To that solution is added dry triethylamine (2.45 ml) and the resulting mixture is cooled to 8-10°. A solution of 6-hydroxy-4,7-dimethoxy-5-benzofurancarboxylic acid methyl ester (Example 6, 250 g) in dry benzene (15 ml) is then added dropwise to the cold reaction mixture over a 15 min
25 period. The resulting mixture is yellow; the ice bath is removed and stirring continued at ambient temperature for 6.5 hours.

B. The reaction mixture is then cooled in ice and quenched by
30 the addition of saturated ammonium chloride (10 ml). Diethyl ether (40 ml) is next added, along with 2N HCl (30 ml). The layers are separated, the ether layer dried (MgSO₄) and concentrated to an oil. This crude mixture was heated with 10 ml of 5% aqueous potassium hydroxide for 2 hr. The mixture is then carefully acidified (6N HCl)
35 and extracted with ethyl acetate (3 x 25 ml). The combined organic extracts are washed with saturated sodium bicarbonate (2 x 20 ml) and dried (MgSO₄). Evaporation of the solvent yields 0.13 g of yellow solid which is chromatographed (on silica gel eluting with 20% ethyl

acetate in Skellysolve B SS8 eluent) to afford 0.128 g of title product. Recrystallization of that material (hexane/ethyl acetate, 10:1) yields 85 mg of pure, bright yellow title product as a solid.

The product of Example 7 is identical to the product of Example 1 of United States Patent 4,284,569 and is accordingly useful for the preparation of khellin and analogs thereof, e.g., see Examples 2-21 of United States Patent 4,284,569.

Example 8: 7-methylthiomethyl-4,9-diethoxyfurochromone (Formula XLIII: R₁ is ethyl, R₁₂ is methylthiomethyl, and R₁₃ is methyl).

Refer to Chart C.

A. 6-Formyl-4,7-dihydroxy-5-benzofurancarboxylic acid methyl ester (Example 4, 4.70 g) is added to acetone (100 ml) followed by addition of ethyl iodide (5.80 g) and potassium carbonate (5.0 g). The resulting mixture is heated at reflux for 24 hr. The reaction is cooled to room temperature and trichloromethane (100 ml) is added. Water (200 ml) is added. The organic layer is separated and the aqueous layer back extracted with trichloromethane (2 x 75 ml). The combined organic layer is dried (MgSO₄) and solvent removed in vacuo to yield a residue. Chromatography over 100 g of silica gel eluting with 5% ethyl acetate in trichloromethane affords product.

B. A solution of 6-Formyl-4,7-diethoxy-5-benzofurancarboxylic acid, methyl ester (part A, 105.0 mg) in isopropanol (8.0 ml) is treated with 85% m-chloroperbenzoic acid (m-CPBA, 188 mg) at ambient temperature and stirred overnight. The solvent is removed on the rotary evaporator and the residue is taken up into 10% aqueous sodium carbonate (10 ml) and diethyl ether (10 ml). After stirring for 30 minutes the layers are separated and the aqueous is extracted with additional diethyl ether (1 x 20 ml). The ether extracts are combined and dried (MgSO₄). The residue is chromatographed (20 g silica gel) eluting with 20% ethyl acetate in isomeric hexanes (Skellysolve B) to provide product.

C. A 100 ml 3 neck flask is oven dried and cooled under nitrogen. Benzene (10 ml) is placed in the flask followed by methyl magnesium bromide (2.9 M in diethyl ether, 2.0 ml). To that solution is added dry triethylamine (2.45 ml) and the resulting mixture is cooled to 8-10°. A solution of 6-hydroxy-4,7-dimethoxy-5-benzofurancarboxylic acid methyl ester (part B, 250 g) in dry benzene (15 ml) is then added

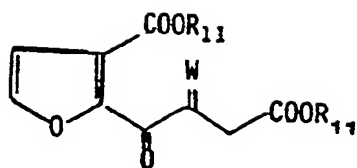
dropwise to the cold reaction mixture over a 15 min period. The ice bath is removed and stirring continued at ambient temperature for 6.5 hours.

5 D. The reaction mixture of part C is then cooled in ice and quenched by the addition of saturated ammonium chloride (10 ml). Diethyl ether (40 ml) is next added, along with 2N HCl (30 ml). The layers are separated, the ether layer dried (MgSO_4) and concentrated to an oil. This crude mixture was heated with 10 ml of 5% aqueous potassium hydroxide for 2 hr. The mixture is then carefully acidified
10 (6N HCl) and extracted with ethyl acetate (3 x 25 ml). The combined organic extracts are washed with saturated sodium bicarbonate (2 x 20 ml) and dried (MgSO_4). Evaporation of the solvent yields a residue which is chromatographed (on silica gel eluting with 20% ethyl acetate in Skellysolve B eluent) to afford product.

15 E. To sodium hydride (20.1 g of a 50% dispersion in oil) and tetrahydrofuran (20 ml freshly distilled from lithium aluminum hydride), combined under a nitrogen atmosphere to form a slurry, are added dropwise a mixture of the product of part D (56 g), ethyl 2-(methylthio)-acetate (26.4 g) and dry tetrahydrofuran (50 ml).
20 After the addition is complete (1.5 hr) the reaction mixture is then heated on a steam bath for 15 min and cooled to ambient temperature. Thereupon excess sodium hydride is destroyed by careful addition of ice and water (300 ml). Washing with diethyl ether (600 ml) yields an aqueous layer which is diluted with methanol (100 ml) and concentrated
25 hydrochloric acid (75 ml). This mixture is then refluxed for 45 min and thereupon allowed to cool to ambient temperature. Upon extraction with methylene chloride (600 ml) the organic extracts are dried and concentrated under reduced pressure to yield pure title product.

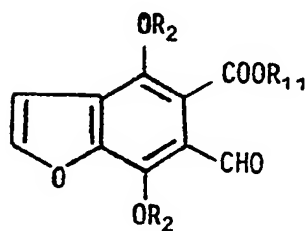
FORMULAS

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I

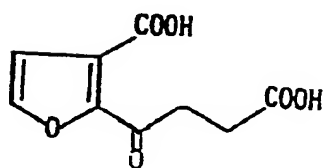
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II

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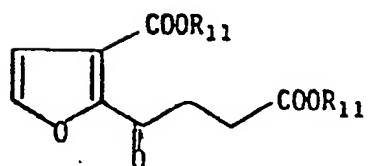
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III

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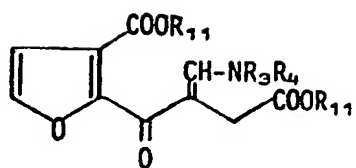
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IV

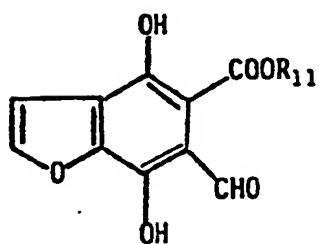
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V

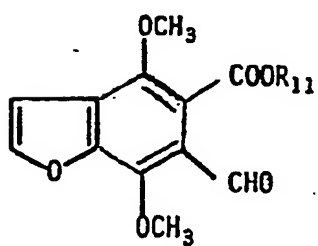
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VI

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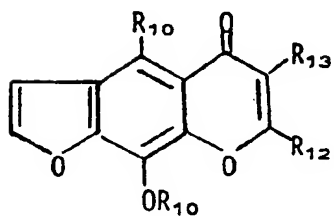
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VII

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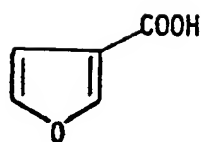


VIII

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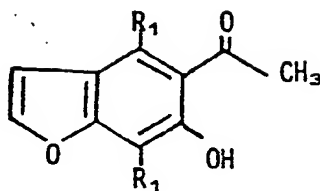
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X

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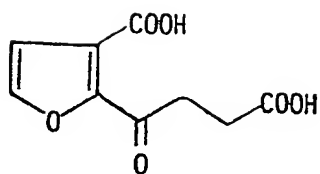
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XI

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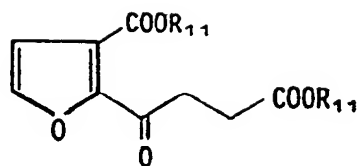


XII

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FORMULAS

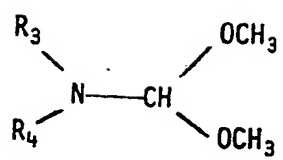
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XIII

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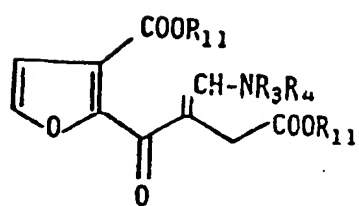
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XIV

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XV

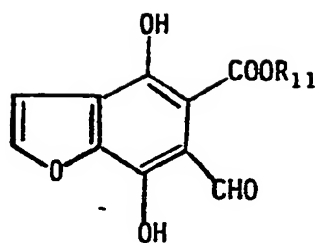
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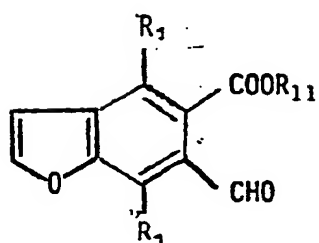
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XVI

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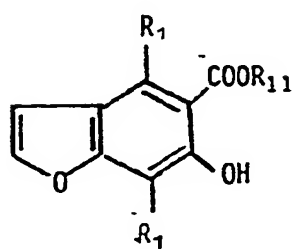
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XVII

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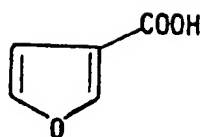


XVIII

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CHART A

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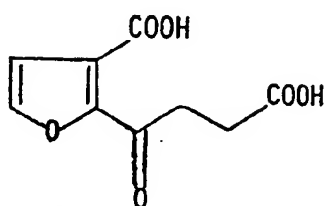


XXI

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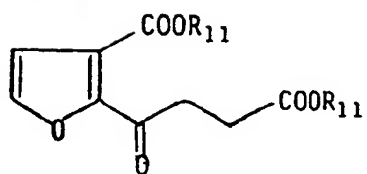


XXII

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XXIII

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to XXIV

CHART A (cont'd.)

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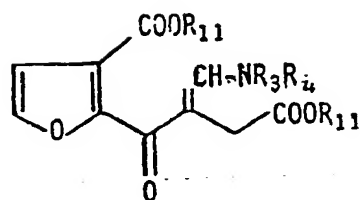
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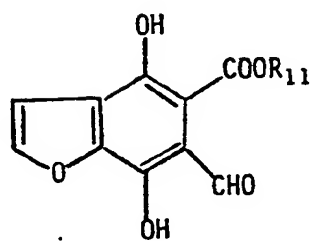
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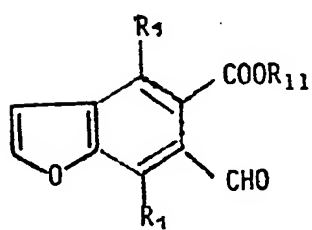
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XXIV



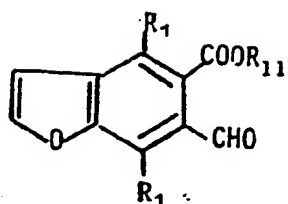
XXV



XXVI

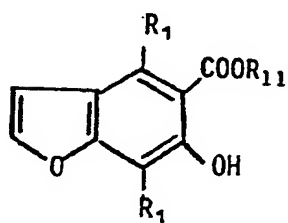
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XXXI

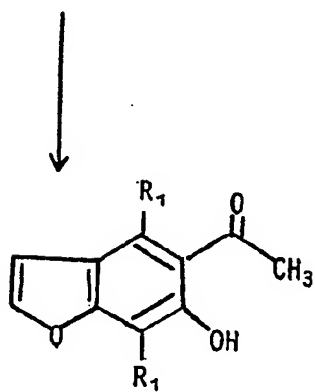
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XXXII

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XXXIII

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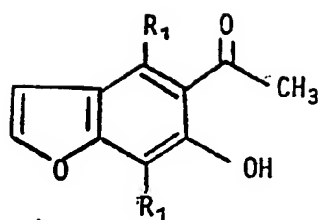
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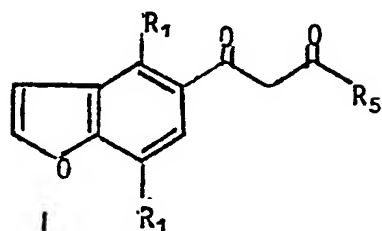
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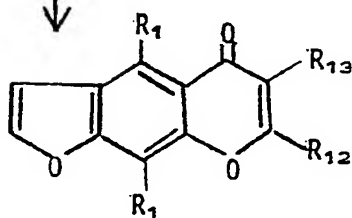
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XLI



XLII



XLIII

APPENDIX BDESCRIPTIONBACKGROUND OF THE INVENTION

5 The present invention provides novel compositions of matter and processes for their preparation. Particularly, the present invention relates to novel chemical intermediates and associated processes for the preparation of furochromones. Most especially, the present invention provides for the preparation of novel antiatherosclerotic furochromones, particularly khellin analogs.

10 Khellin and related compounds are known to exert a wide variety of pharmacological effects. Recently, khellin has been reported to exhibit useful antiatherosclerotic activities. Moreover, numerous analogs of khellin likewise are known to exert useful antiatherosclerotic effects. For example, 7-methylthiomethyl-4,9-dimethoxy-
15 furochromone is described in United States Patent 4,284,569 as such a useful antiatherosclerotic substance.

Methods for the total synthesis of khellin are known. For example, pyrogallol has been employed as a starting material for the synthesis of furochromones such as khellin. See Clarke, J.R., et al.,
20 J. Chem. Soc., 302 (1949), Baxter, R.A., et al., J. Chem. Soc., 530 (1949), Schonberg, A., et al., J. Am. Chem. Soc., 73:2960 (1951), Murti, V.V.S., et al., Proc. of the Indian Acad. of Sci., 30A:107 (1949), and Geissman, T.A., et al., J. Am. Chem. Soc., 73:1280 (1951). Also descriptive of the synthesis of khellin are Spath, E., et al.,
25 Chem. Ber., 71:106 (1938), Dann, O., et al., Chem. Ber., 93:2829 (1960), Dann, O., et al., Ann. Chem., 605:146 (1957), and Murti, V.V.S., et al., J. Sci. Ind. Res. (India), 8B:112 (1949). See also United States Patent 2,680,119 describing the synthesis of khellin and related compounds.

30 Other references describing the synthesis of intermediates useful in the preparation of khellin for analogs include: Aneja, R., et al., Chem. Ber., 93:297 (1960), Aneja, R., et al., J. Sci. Ind. Res. (India), 17B:382 (1958), Gardner, T.S., et al., J. Org. Chem., 15:841 (1950), and Rowe, L.R., et al., Indian J. Chem., 5:105 (1967).

35 Accordingly, the references cited above describe the preparation of 1-(6-hydroxy-4,7-dimethoxy-5-benzofuranyl)-ethanone. Also known is the related compound 6-hydroxy-4,7-dimethoxy-5-benzofurancarboxylic acid, methyl ester, described by Musante, C., Gazz. Chim. Ital.,

88:910 (1958).

PRIOR ART

Methods of the total synthesis of khellin are known, as are certain chemical intermediates useful in its synthesis.

- 5 The use of pyrogallol in the synthesis of khellin intermediates is known. For example, the transformation of pyrogallol to the khellin intermediate 1-(2,3-dihydro-6,7-dihydroxy-5-benzofuranyl) ethanone is known. The parahydroxylation of this intermediate is also known. See Row, L.R., et al., Indian J. Chem., 5:105 (1967)
10 describing this transformation and the subsequent dimethylation to yield known khellin intermediates. United States Patent 4,284,569 provides a variety of novel anti-atherosclerotic furochromones.

SUMMARY OF THE INVENTION

The present invention particularly provides:

- 15 (a) A method of preparing a dialkoxybenzofuran of formula I, wherein R_2 and R_3 are C_1 - C_4 alkyl, being the same or different, which comprises para-alkoxylating a mono-alkoxybenzofuran of formula II, wherein R_2 is as defined above, with an oxidizing reagent selected from the group consisting of
20 (i) thallium (III) nitrate, (ii) ceric ammonium nitrate, (iii) lead tetraacetate,
in a C_1 - C_4 alkanol solvent of the formula R_3OH ,
wherein R_3 is as defined above;
 (b) A benzofuran of formula III,
25 wherein R_2 is C_1 - C_4 alkyl;
 (c) A benzofuran of formula IV,
wherein R_5 is C_2 - C_4 alkyl;
 (d) A benzofuran of formula V,
wherein one of R_6 and R_7 is C_1 - C_4 alkyl and the other is C_2 - C_4
30 alkyl with the proviso that R_6 and R_7 are different;
 (e) An anti-atherosclerotic furochromone of formula VI,
wherein R_6 and R_7 are as defined above;
wherein R_{12} is:
 (1) hydrogen;
35 (2) C_1 - C_8 alkyl;
 (3) C_2 - C_8 alkoxymethyl;
 (4) C_2 - C_8 alkylthioalkyl;
 (5) trifluoromethyl;

(6) phenoxymethyl optionally substituted by chloro, fluoro, trifluoromethyl, C₁-C₃ alkyl or C₁-C₃ alkoxy;

(7) phenylthiomethyl optionally substituted by chloro, fluoro, trifluoromethyl, C₁-C₃ alkyl or C₁-C₃ alkoxy;

5 (8) -CH₂-S(O)_n-R₂₀, wherein n is zero, one or 2 and R₂₀ is C₁-C₅ alkyl; or

(9) -CH₂NR₈R₉, wherein R₈ and R₉ are hydrogen, C₁-C₁₂ alkyl or wherein R₈ and R₉, taken together with N, form a saturated or unsaturated heterocyclic amine ring consisting of from 2 to 7 carbon
10 atoms, inclusive, and zero, one, or 2 additional hetero atoms, with the proviso that said heterocyclic amine ring contains 4 to 8 atoms in the ring, said additional hetero atoms being selected from the group consisting of oxygen, nitrogen, and sulfur, said heterocyclic amine ring being optionally substituted by C₁-C₄ alkyl, C₂-C₈ alkylthio-
15 methyl or alkoxymethyl C₁-C₄ hydroxyalkyl, or phenyl;

wherein R₁₃ is:

(1) hydrogen;

(2) chloro, iodo, or bromo; or

(3) -CH₂-S(O)_n-R₂₀ wherein n and R₂₀ are as defined above,

20 with the proviso that R₁₃ is -CH₂-S(O)_n-R₂₀ only when R₁₄ is methyl;

(f) 4-Ethoxy-9-methoxy-7-methylthiomethylfurochromone; and

(g) 4-Methoxy-9-ethoxy-7-methylthiomethylfurochromone.

In accordance with the method described above, there is prepared the formula II alkoxybenzofuran. This formula II alkoxybenzofuran
25 wherein R₂ is methyl is known to be useful in the preparation of a wide variety of anti-atherosclerotic substances, including khellin and various analogs thereof. See United States Patent 4,284,569.

Similarly there are prepared the novel formula XI benzofurans when R₁ is C₂-C₄ alkoxy. These intermediates are useful in the
30 preparation of novel anti-atherosclerotic 4,9-di-(C₂-C₄)-alkoxy-furochromones of formula VIII by means described in United States Patent 4,284,569 for the preparation of the corresponding 4,9-dimethoxyfurochromones therein. Moreover, the manner of use of the novel 4,9-di-(C₂-C₄)-alkoxy-furochromones of formula VIII in the
35 treatment and prevention of atherosclerosis is the same as that described in United States Patent 4,284,569 for the corresponding 4,9-dimethoxy compounds. Accordingly, the manner of the preparation and pharmacological use of these novel formula VIII compounds is

incorporated herein by and reference from the description of the preparation and use in United States Patent 4,284,569 of the anti-atherosclerotic 4,9-dimethoxyfurochromones. Among the novel formula VIII compounds herein, the 4,9-diethoxyfurochromones are preferred.

5 The process of the present invention is more completely understood by reference to the charts below. In these charts, R_1 is hydrogen or C_1 - C_4 alkyloxy or hydrogen. R_2 , R_3 , R_{12} and R_{13} are as defined above.

R_{11} is:

- 10 (a) hydrogen;
 (b) C_1 - C_8 alkyl;
 (c) C_2 - C_8 alkoxyethyl;
 (d) C_2 - C_8 alkylthioalkyl;
 (e) trifluoromethyl;
15 (f) phenoxyethyl;
 (g) phenylthiomethyl;
 (h) phenoxyethyl or phenylthiomethyl, either of which is optionally substituted by one chloro, fluoro, trifluoromethyl, C_1 - C_3 -alkyl, or C_1 - C_3 -alkoxy; or
20 (i) C_3 - C_{10} cycloalkyl.

The carbon atom content of various hydrocarbon containing moieties is indicated by a prefix designating the minimum and maximum number of carbon atoms in the moiety, i.e., the prefix C_i - C_j indicates a carbon atom content of the integer "i" to the integer "j" carbon atoms, inclusive. Thus, C_1 - C_3 alkyl refers to alkyl of 1-3 carbon atoms, inclusive, or methyl, ethyl, propyl, and isopropyl.

25 With respect to the charts, Chart A provides a method whereby the known formula XXI compound is transformed to the novel formula XXV 2,3-dihydro-benzofuran, which is in turn reduced to the known formula XXVI benzofuran intermediate for preparing desmethoxy khellin and other khellin analogs.

30 With further respect to Chart A, pyrogallol is converted to the formula XXII triacetate first by treatment with chloroacetonitrile according to procedures described by Geissman, T.A., et al., J. Amer. Chem. Soc., 73:57-65 (1951).

35 Thereafter the formula XXIII production product is obtained from the formula XXII compound employing a metal catalyst under a hydrogen atmosphere. For example, conventional metal catalysts such as

palladium and carbon catalysts are employed. See Dann, O. and Zeller, H.G., Ber. 93:28-29 (1960) for a transcription of this transformation.

Thereafter the formula XXIII compound then goes to Fries rearrangement to yield formula XXIV dihydroxy ketone. By this
5 procedure, the formula XXIII compound is treated with a mixture of aluminium trichloride and nitrobenzene.

This formula XXIV compound is then selectively C₁-C₄ alkylated to the formula XXV compound by treatment with an alkyl iodide. Any concomitant desalkoxylation in this reaction is reversed by treatment
10 with hydrobromic acid.

This novel formula XXV compound is then dehydrogenated using 2,3-dichloro-5,6-dicyanobenzoquinone (DDQ). See Linn, Y.Y., et al., J. Heterocyclic Chem., 799 (1979).

The formula XXVI compound thusly prepared is a highly useful
15 intermediate in the synthesis of analogs of khellin, specifically the 4-desmethoxy khellin. See, for example, United States Patent 4,284,569, describing the synthesis of such analogs from the formula XXVI compound wherein R₂ is methyl.

With respect to Chart C, a method is provided for the para-
20 alkoxylation of the formula XXXI compound, prepared as the formula XXVI compound in Chart A. This methoxylation proceeds by the use of an oxidizing agent in a C₁-C₄ alkanol solvent corresponding to the alkoxy group to be introduced at C₄. Ce(NH₄)₂(NO₃)₆ or Pb(OAc)₄ may be employed as the oxidizing agent. See Brother, A.E., et al.,
25 Helv. Chim. Acta., 35:9-10 (1952). Alternatively, however, thallium (III) nitrate is employed as the oxidizing agent. See Taylor, E.C., J. Organic Chem., 41:282 (1976). In the latter case, maximum yields are obtained when the oxidizing agent is added over a period of about 15 min with the reaction mixture being maintained at about -25°C for
30 30 min followed by heating for 1-2 min.

Chart C provides an illustration of the method by which the formula XLI compound, prepared as the formula XXVI compound of Chart A or the formula XXXII compound of Chart B, is transformed to khellin 4-desmethoxy khellin or analogs thereof. Procedures of Chart C are, for
35 example, known in the art from United States Patent 4,284,569 wherein Chart A of that patent describes the synthesis of the various formula XLII and formula XLIII compounds from the formula XLI starting material. Accordingly, the charts herein provide a description of the

preparation and use of the novel process and compounds of the present invention.

DESCRIPTION OF THE PREFERRED EMBODIMENTS

The present disclosure is more fully understood by the operation
5 of the following examples:

Example 1: 3,6,7-Benzofurantriol triacetate (Formula XXII).

Refer to Chart A.

Dry zinc chloride (38 g) is added to pyrogallol (Formula XXII, 35 g). To these solids under a nitrogen atmosphere is added diethyl
10 ether and chloroacetonitrile (18 ml). The resulting mixture is then stirred and cooled to 0°C. Hydrochloride gas is then bubbled to the reaction mixture for 30 min, the mixture is allowed to warm slowly to ambient temperature with stirring for 12 hr. Thereafter, the two-face mixture is cooled to 0°C and the ether layer decanted. Additional
15 diethyl ether (100 mg) is added, stirred, and decanted. Water (250 ml) is added to the resulting residue and the aqueous mixture is then refluxed for 30 min yielding a homogeneous solution. The solution is then cooled to 4°C, filtered, and yields a reddish-brown solid (24.5 g), α -chlorogallacetophenone. Without further purification, the solid
20 is then dissolved in ethanol (300 ml) containing sodium acetate (24.5 g). After refluxing for 5 hr, the resulting mixture is then dried and treated with acetic anhydride (150 ml) and pyridine (75 ml). The resulting mixture is then stirred at ambient temperature for 12 hr, decanted into ice water (700 ml) and stirred for one hr. The
25 resulting precipitate is then collected on a filter, washed with water and air dried. Thereafter, the filtrate is acidified with co-concentrated hydrochloric acid and extracted with ethyl acetate. The organic extracts are then washed thoroughly with saturated sodium bicarbonated brine, dried over magnesium sulphate, and concentrated to
30 a residue. Chromatography on 1.3 kg of silica gel alluding with 40% ethyl acetate Skellysolve B ethyl acetate to obtain 27.75 g of pure product, a white solid, melting point 99-101°C.

Example 2: 2,3-Dihydro-6,7-benzofurandiol diacetate (Formula XXIII).

35 A mixture of the title product of Example 1 (50 g) in ethyl acetate (350 ml) is treated with anhydrous potassium acetate (7.5 g) and 7.5 g of a 10% paladium on carbon catalyst. The reaction mixture is then hydrogenated at 65°C for 2 hr, cooled to ambient temperature,

and filtered through diatomaceous earth. The resulting filtrate is then concentrated under reduced pressure yielding a solid. Recrystallization from 250 ml of ethyl acetate and Skellysolve B (1:1) yields 33.65 g of pure title product as a white solid, melting point 114-115°C. Silica gel TLC R_f is 0.31 in 25% ethyl acetate in hexane. IR absorptions (cm^{-1}) are observed at 1765, 1625, 1610, 1490, 1465, 1375, 1225, 1210, 1180, and 1035. NMR-absorptions are observed at 7.05, 6.62, 4.64, 3.21, 2.25, and 2.23 δ .

5 Example 3: 1-(2,3-dihydro-6,7-dihydroxy-5-benzofuranyl)-ethanone (Formula XXIV).

10 Refer to Chart A.

A mixture of the title product of Example 2 (9.50 g), nitrobenzene (100 ml) and aluminium trichloride (6.36 g) is heated at 60°C for 90 min. After cooling to ambient temperature, the reaction mixture is poured over ice and 2 N hydrochloric acid (100 ml) is added, followed by the addition of water (300 ml). After stirring for 3 hr, the reaction mixture is then extracted with ethyl acetate. The organic layer is then separated and washed with 5% aqueous sodium hydroxide and the aqueous layer then poured into 2 N hydrochloric acid (500 ml), yielding a precipitate. Ethyl acetate is then added and then separated from the aqueous layer. The organic layer is then dried over magnesium sulphate and concentrated under reduced pressure to yield 6.35 g of title product as a brown solid. Recrystallization from ethyl acetate yields pure crystalline product, melting point 190-190.5°C. Silica gel TLC R_f is 0.25 in 5% ethyl acetate in trichloromethane. IR absorptions (cm^{-1}) are observed at 3460, 3200, 1645, 1605, 1490, 1445, 1365, 1320, 1255, and 1055. NMR absorptions are observed at 7.15, 4.70, 3.18, and 2.52 δ .

25 Example 4: 1-(2,3-dihydro-6-hydroxy-7-methoxy-5-benzofuranyl)-ethanone (Formula XXV: R_2 is methyl).

30

Refer to Chart A.

A mixture of the title product of Example 3 (5.9 g) potassium carbonate (12 g) and methyl iodide (25 g) is heated at reflux and acetone for 18 hr. After cooling to ambient temperature and removal of the potassium carbonate by filtration, the resulting mixture is then concentrated under reduced pressure to a yellow oil. The oil is then dissolved in trichloromethane and treated with hydrobromic acid and refluxed for 2 hr. After cooling to ambient temperature and

35

concentration under reduced pressure, chromatography eluting with 5% ethyl acetate in trichloromethane yields 6.0 g of pure title product, melting point 95-97°C. Silica gel TLC R_f is 0.5 in 5% ethyl acetate in trichloromethane. IR absorptions (CM^{-1}) are observed at 2700, 1630, 1615, 1430, 1405, 1365, 1330, 1290, and 1060. NMR absorptions are observed at 7.3, 6.48, 3.9, and 3.15 δ .

Example 5: 1-(6-hydroxy-7-methoxy-5-benzofuranyl)-ethanone
(Formula XXVI: R_2 is methyl).

Refer to Chart A.

10 To a solution of the title product of Example 4 (12.1 g) in dioxane is added 2,3-dichloro-5,6-dicyanobenzoquinone (DDQ, 17.8 g). The resulting black solution is then stirred at reflux for 2 hr. Upon cooling to ambient temperature, a precipitate forms. The reaction mixture is then filtered and the filter cake is washed with dichloro-
15 methane. The filtrate is then concentrated to a residue and the residue chromatographed on 700 g of silica gel eluting with dichloromethane. Pure title product is obtained as 5.5 g of an oil which spontaneously crystallizes. Recrystallization from ethyl acetate in hexane (1:5) yields pure title product, melting point 61.2-63.5°C.
20 Silica gel TLC R_f is 0.31 in 25% ethyl acetate in hexane. IR absorptions (CM^{-1}) are observed at 3140, 3110, 2720, 1635, 1625, 1545, 1320, 1300, 1275, and 1050. NMR absorptions are observed at 7.78, 7.65, 4.21, and 2.71 δ .

Example 6: 1-(6-hydroxy-4,7-dimethoxy-5-benzofuranyl)-ethanone
25 (Formula XXXII: R_2 and R_3 are both methyl).

Refer to Chart B.

A. The title product of Example 5 (100 mg) is added to methanol (4 ml) and cooled to -25°C. To the resulting heterogeneous mixture is added a methanolic (7 ml) solution of thallium (III) nitrate trihydrate, $\text{TL}(\text{ONO}_2)_3 \cdot 3\text{H}_2\text{O}$ (250 mg), dropwise over about 15 min. The resulting mixture is then stirred for 30 min at -25°C and heated to reflux for 1-2 min. The reaction is then poured into saturated aqueous sodium bicarbonate and extracted with diethyl ether. The ethereal layer is then dried over magnesium sulphate and concentrated
30 under reduced pressure to yield a yellow oil. Crystallization is achieved by dissolving the oil in 1% ethyl acetate in hexane and cooling to 0°C for 12 hr. Filtration of the resulting crystals yields 70 mg of pure title product, melting point 98-99°C. Silica gel TLC R_f

is 0.6 in ethyl acetate at hexane (1:1). IR absorptions (CM^{-1}) are observed at 2955, 2930, 2926, 2868, 1629, 1619, 1587, 1471, 1452, 1444, 1425, 1382, 1364, 1303, 1267, 1151, 1079, 1061, and 755. NMR absorptions are observed at 7.5, 6.9, 4.15, 4.05, and 2.7 δ .

- 5 B. Alternatively title product is prepared utilizing lead tetraacetate (200 mg) which is added to methanol (6 ml) and cooled to 0°C. To the resulting solution is added the title product of Example 5 (100 mg) dropwise in methanol (5 ml). The resulting mixture is then stirred at 0°C for 80 min and poured into saturated aqueous sodium
10 bicarbonate. After extraction with ether, the ethereal solution is then dried over magnesium sulphate and concentrated under reduced pressure to yield a yellow oil. This crude solid is then dissolved in methanol and heated at reflux for 1 hr. After cooling to ambient temperature and removal of solvent under reduced pressure, crystalli-
15 zation from 1% ethyl acetate in hexane yields 70 mg of pure title product, melting point 98-100°C.

Example 7: 7-Methylthiomethyl-7-methoxy-9-ethoxy-furo-
chromone (Formula XXIII: R_1 is methyl, R_2 is ethyl,
 R_{12} is methylthiomethyl, and R_{13} is hydrogen).

- 20 Refer to Charts A, B, and C.

A. A mixture of the title product of Example 3 (5.9 g) potassium carbonate (12 g) and ethyl iodide (28 g) is heated at reflux and acetone for 18 hr. After cooling to ambient temperature and removal of the potassium carbonate by filtration, the resulting mixture is
25 then concentrated under reduced pressure. The residue is then dissolved in trichloromethane and treated with hydrobromic acid and refluxed for 2 hr. After cooling to ambient temperature and concentration under reduced pressure, chromatography eluting with 5% ethyl acetate in trichloromethane yields 6.0 g of formula XXV product.

30 B. To a solution of the product of Part A (12 g) in dioxane is added 2,3-dichloro-5,6-dicyanobenzoquinone (DDQ, 17.8 g). The resulting black solution is then stirred at reflux for 2 hr. Upon cooling to ambient temperature, a precipitate forms. The reaction mixture is then filtered and the filter cake is washed with dichloromethane. The
35 filtrate is then concentrated to a residue and the residue chromatographed on 700 g of silica gel eluting with dichloromethane to obtain formula XXVI product, 1-(6-hydroxy-7-ethoxy-5-benzofuranyl)-ethanone.

C. The formula XXVI product of Part B (100 mg) is added to

methanol (4 ml) and cooled to -25°C. To the resulting mixture is added a methanolic (7 ml) solution of thallium (III) nitrate trihydrate, TL $(\text{ONO}_2)_3 \cdot 3\text{H}_2\text{O}$ (250 mg), dropwise over about 15 min. The resulting mixture is then stirred for 30 min at -25°C and heated to reflux for 1-2 min. The reaction is then poured into saturated aqueous sodium bicarbonate and extracted with diethyl ether. The ethereal layer is then dried over magnesium sulphate and concentrated under reduced pressure to yield formula XXXII product, 1-(6-hydroxy-4-methoxy-7-ethoxy-5-benzofuranyl)-ethanone.

10 D. To sodium hydride (20.1 g of a 50% dispersion in oil) and tetrahydrofuran (20 ml freshly distilled from lithium aluminum hydride), combined under a nitrogen atmosphere to form a slurry, are added dropwise a mixture of the product of Part C (56 g), ethyl 2-(methylthio)-acetate (26.4 g) and dry tetrahydrofuran (50 ml). After 15 the addition is complete (1.5 hr) the reaction mixture is then heated on a steam bath for 15 min and cooled to ambient temperature. Thereupon excess sodium hydride is destroyed by careful addition of ice and water (300 ml). Washing with diethyl ether (600 ml) yields an aqueous layer which is diluted with methanol (100 ml) and concentrated 20 hydrochloric acid (75 ml). This mixture is then refluxed for 45 min and thereupon allowed to cool to ambient temperature. Upon extraction with methylene chloride (600 ml) the organic extracts are dried and concentrated under reduced pressure to yield pure title product.

25 Example 8: 7-Methylthiomethyl-4-ethoxy-9-methoxy-furochromone (Formula XXIII: R_1 is ethyl, R_2 is methyl, R_{12} is methylthiomethyl, and R_{13} is hydrogen.)

Refer to Chart A, B, and C.

A. The title product of Example 5 (100 mg) is added to ethanol (4.5 ml) and cooled to -25°C. To the resulting mixture is added an 30 ethanolic (8 ml) solution of thallium (III) nitrate trihydrate, TL $(\text{ONO}_2)_3 \cdot 3\text{H}_2\text{O}$ (250 mg), dropwise over about 15 min. The resulting mixture is then stirred for 30 min at -25°C and heated to reflux for 1-2 min. The reaction is then poured into saturated aqueous sodium bicarbonate and extracted with diethyl ether. The ethereal layer is 35 then dried over magnesium sulphate and concentrated under reduced pressure to yield formula XXXII product, 1-(6-hydroxy-4-ethoxy-7-methoxy-5-benzofuranyl)-ethanone.

B. To sodium hydride (20.1 g of a 50% dispersion in oil) and

tetrahydrofuran (20 ml freshly distilled from lithium aluminum hydride), combined under a nitrogen atmosphere to form a slurry, are added dropwise a mixture of the product of Part A (56 g), ethyl 2-(methylthio)-acetate (26.4 g) and dry tetrahydrofuran (50 ml). After the addition is complete (1.5 hr) the reaction mixture is then heated on a steam bath for 15 min and cooled to ambient temperature. Thereupon excess sodium hydride is destroyed by careful addition of ice and water (300 ml). Washing with diethyl ether (600 ml) yields an aqueous layer which is diluted with methanol (100 ml) and concentrated hydrochloric acid (75 ml). This mixture is then refluxed for 45 min and thereupon allowed to cool to ambient temperature. Upon extraction with methylene chloride (600 ml) the organic extracts are dried and concentrated under reduced pressure to yield pure title product.

C. Alternatively title product is prepared as follows:

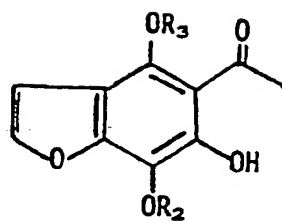
(1) 4,7-Dimethoxy-7-[(methylthio)methyl]-furochromone (15 g) is added to trichloromethane (250 ml). Anhydrous hydrobromic acid is then bubbled through the resulting mixture until a dark red color develops. The reaction is then heated to reflux for 45 min, cooled to ambient temperature, and diluted with water (200 ml). The organic layer is then separated, dried over magnesium sulphate, and concentrated under reduced pressure to yield 13.36 g of 4-hydroxy-7-[(methylthio)methyl]-9-methoxy-furochromone. Melting point 134-135° C.

(2) The product of part C(1) (4.0 g) is added to acetone (100 ml), ethyl iodide (15 ml) and potassium carbonate (9 g). The resulting mixture is then heated to reflux for 18 hr, cooled to ambient temperature, and concentrated under reduced pressure. The resulting solid is then washed with trichloromethane and separated by filtration. Concentration under reduced pressure yields a dark oil which is chromatographed on 300 gr of silica gel by high pressure liquid chromatography. Packing in elution with 10% ethyl acetate in trichloromethane yields 3.0 g of title product, melting point 112-114° C. Silica gel TLC Rf is 0.78 in 1% methanol in ethyl acetate. IR absorptions (cm^{-1}) are observed at 3120, 1650, 1610, 1380, 1340, 1210, 1170, and 1065. NMR absorptions are observed at 7.62, 6.97, 6.15, 4.21, 4.20, 4.57, and 2.21 δ .

FORMULAS

5

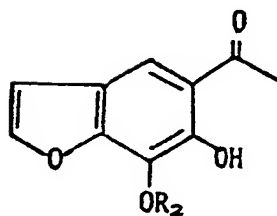
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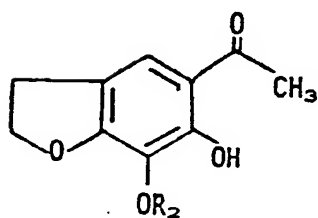
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II

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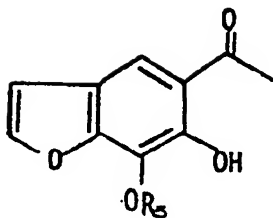
III

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FORMULAS (cont'd)

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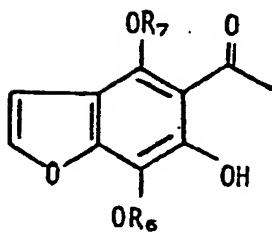
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IV

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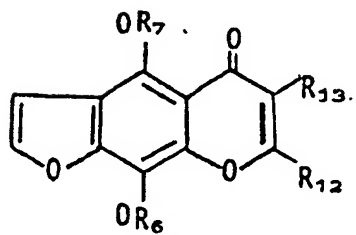
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V

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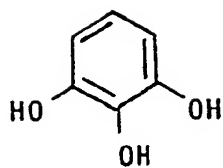


VI

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CHART A

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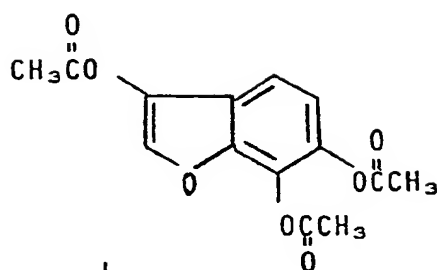


XXI

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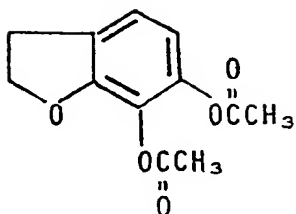


XXII

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XXIII

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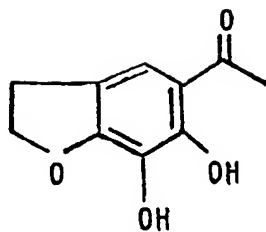


to XXIV

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CHART A (cont'd)

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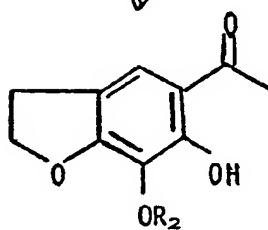


XXIV

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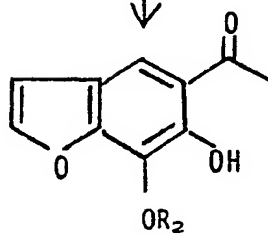


XXV

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XXVI

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CHART B

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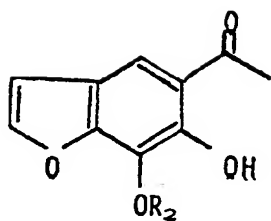
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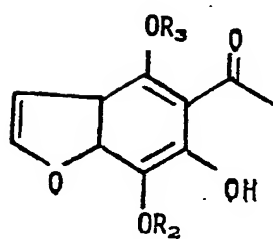
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XXXI



XXXII

CHART C

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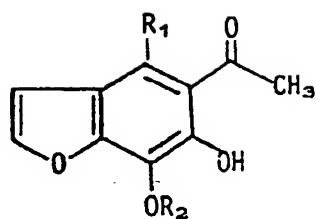
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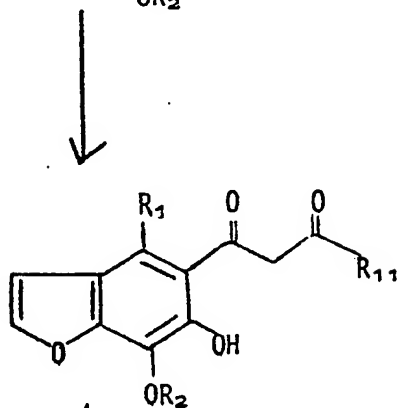
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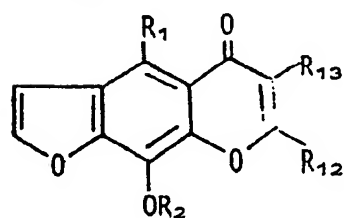
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XL I



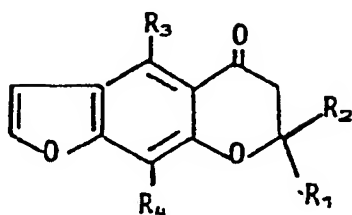
XL II



XL III

CLAIMS

1. A dihydrofurochromone of formula I:



I

wherein R_1 and R_2 , being the same or different, are individually:

- (a) C_1 - C_6 alkyl,
- (b) trifluoromethyl,
- (c) C_5 - C_{10} cycloalkyl with the proviso that the cycloalkyl ring is C_5 - C_7 ,
- (d) C_2 - C_8 alkylaminoalkyl,
- (e) C_2 - C_8 alkoxyalkyl,
- (f) C_2 - C_8 alkylthioalkyl,
- (g) C_2 - C_8 alkylsulfinylalkyl,
- (h) C_2 - C_8 alkylsulfonylalkyl,
- (i) C_7 - C_{12} phenoxyalkyl optionally substituted on the phenyl ring by one, 2, or 3,
 - (i) hydroxy,
 - (ii) C_1 - C_3 alkoxy,
 - (iii) C_1 - C_3 alkyl,
 - (iv) trifluoromethyl,
 - (v) halo which is fluoro, chloro, or bromo, with the proviso that not more than two such substituents are other than alkyl,
- (j) C_7 - C_{12} phenylthioalkyl optionally substituted on the phenyl ring by one, 2, or 3,
 - (i) hydroxy,
 - (ii) C_1 - C_3 alkoxy,
 - (iii) C_1 - C_3 alkyl,
 - (iv) trifluoromethyl,
 - (v) halo which is fluoro, chloro, or bromo, with the proviso that not more than two such substituents are other than alkyl,

- (k) phenyl optionally substituted by one, 2, or 3,
 (i) hydroxy,
 (ii) C₁-C₃ alkoxy,
 (iii) C₁-C₃ alkyl,
5 (iv) trifluoromethyl,
 (v) halo which is fluoro, chloro, or bromo, with the
proviso that not more than two such substituents are other than alkyl,
 (l) aralkyl optionally substituted on the aromatic ring by
one, 2, or 3,
10 (i) hydroxy,
 (ii) C₁-C₃ alkoxy,
 (iii) C₁-C₃ alkyl,
 (iv) trifluoromethyl,
 (v) halo which is fluoro, chloro, or bromo, with the
15 proviso that not more than two such substituents are other than alkyl,
 (m) 2- or 3-furanyl optionally substituted by
 (i) hydroxy,
 (ii) C₁-C₃ alkoxy,
 (iii) C₁-C₃ alkyl,
20 (iv) trifluoromethyl,
 (v) halo which is fluoro, chloro, or bromo, with the
proviso that not more than two such substituents are other than alkyl,
 (n) 2- or 3-thenyl optionally substituted by
 (i) hydroxy,
25 (ii) C₁-C₃ alkoxy,
 (iii) C₁-C₃ alkyl,
 (iv) trifluoromethyl,
 (v) halo which is fluoro, chloro, or bromo, with the
proviso that not more than two such substituents are other than alkyl,
30 or
 (o) -CH₂NR₈R₉ wherein R₈ and R₉, being the same or
different, are individually,
 (i) hydrogen,
 (ii) C₁-C₈ alkyl,
35 (iii) C₅-C₁₀ cycloalkyl,
 (iv) C₇-C₁₂ aralkyl, or
 (v) phenyl optionally substituted by one, 2, or 3
 a) hydroxy,

b) C₁-C₃ alkoxy,

c) C₁-C₃ alkyl,

d) trifluoromethyl,

e) halo which is fluoro, chloro, or bromo, with

5 the proviso that not more than two such substituents are other than alkyl, or wherein R₈ and R₉ are taken together with the nitrogen to form a saturated or unsaturated heterocyclic amine ring consisting of from 2 to 7 carbon atoms, inclusive, and zero, one, or 2 additional hetero atoms or hetero atom groups, with the proviso that said
10 heterocyclic amine ring contains 4 to 8 atoms in the ring, said additional hetero atoms or hetero atom groups being selected from the group consisting of oxygen, nitrogen, sulfur, -SO- and -SO₂- said heterocyclic amine ring being optionally substituted by C₁-C₄ alkyl, C₂-C₈ alkylthiomethyl, or C₂-C₈ alkoxymethyl or C₁-C₄ hydroxymethyl or
15 phenyl; or

wherein R₁ and R₂ are taken together and form a bivalent moiety which is:

(a) -CH₂-(CH₂)_a-CH₂- wherein the integer "a" is zero to 5;

(b) -CH₂-(CH₂)_b-X-(CH₂)_c-CH₂-

20 wherein the integer "b" is zero and the integer "c" is zero, one, 2, or 3 or the integer "b" is one and the integer "c" is zero, one, or 2, and wherein X is oxa (-O-), thia (-S-), or -N(R₁₀)- wherein R₁₀ is

(i) hydrogen,

(ii) C₁-C₈ alkyl,

25 (iii) C₅-C₁₀ cycloalkyl,

(iv) C₇-C₁₂ aralkyl, or

(v) phenyl optionally substituted by one, 2, or 3

a) hydroxy,

b) C₁-C₃ alkoxy,

30 c) C₁-C₃ alkyl,

d) trifluoromethyl,

e) halo which is fluoro, chloro, or bromo, with the

proviso that not more than two such substituents are other than alkyl;

wherein R₃ is hydrogen, or C₁-C₄ alkoxy and R₄ is hydrogen or
35 C₁-C₄ alkoxy, with the provisos that one of R₃ and R₄ is hydrogen only when the other is other than hydrogen and R₃ and R₄ are the same or different.

2. A compound according to claim 1 selected from the group consisting of

- 6,7-Dihydro-4,9-dimethoxy-7,7-dimethyl-5H-furo[3,2-g][1]benzopyran-5-one,
- 5 7-Ethyl-6,7-dihydro-4,9-dimethoxy-7-methyl-5H-furo[3,2-g][1]benzopyran-5-one,
- 7,7-Diethyl-6,7-dihydro-4,9-dimethoxy-5H-furo[3,2-g][1]benzopyran-5-one,
- 6,7-Dihydro-4,9-dimethoxy-7-methyl-7-phenyl-5H-furo[3,2-g][1]benzopyran-5-one,
- 10 4',9'-Dimethoxy-spiro[cyclopentane-1,7'-[7H]furo[3,2-g][1]benzopyran]-5'(6'H)-one,
- 4',9'-Dimethoxy-spiro[cyclohexane-1,7'-[7H]furo[3,2-g][1]benzopyran]-5'(6'H)-one,
- 15 2',3',5',6'-Tetrahydro-4,9-dimethoxy-spiro[7H-furo[3,2-g][1]benzopyran-7,4'-[4H]thiopyran]-5(6H)-one,
- 2',3',5',6'-Tetrahydro-4,9-dimethoxy-spiro[7H-furo[3,2-g][1]benzopyran-7,4'-[4H]thiopyran]-5(6H)-one, 1'-oxide,
- 2',3',5',6'-Tetrahydro-4,9-dimethoxy-spiro[7H-furo[3,2-g][1]benzopyran-7,4'-[4H]thiopyran]-5(6H)-one, 1',1'-dioxide,
- 20 6,7-Dihydro-4,9-dimethoxy-7-methyl-7-[(methylthio)methyl]-5H-furo[3,2-g][1]benzopyran-5-one,
- 6,7-Dihydro-4,9-dimethoxy-7-methyl-7-[(methylthio)methyl]-5H-furo[3,2-g][1]benzopyran-5-one, oxide,
- 25 6,7-Dihydro-4,9-dimethoxy-7-methyl-7-[(methylthio)methyl]-5H-furo[3,2-g][1]benzopyran-5-one, dioxide, and
- 4,9-Dimethoxy-1'-methyl-spiro[7H-furo[3,2-g][1]benzopyran-7,4'-piperidine]-5(6H)-one.



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EUROPEAN SEARCH REPORT

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EP 83 30 2549

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. Cl. 3)
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A	US-A-4 304 722 (UPJOHN)		A 61 K 31/38 A 61 K 31/35 (C 07 D 493/04 C 07 D 311/00 C 07 D 307/00)
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D, A	US-A-4 284 569 (UPJOHN)		(C 07 D 491/20 C 07 D 311/00 C 07 D 307/00) C 07 D 221/00)

A	Chemical Abstracts vol. 90, no. 23, 4 June 1979, Columbus, Ohio, USA H. ABU-SHADY et al. "Experiments with khellin. VIII: Synthesis of some 2-3-substituted furochromones", page 653, column 2, abstract no. 186837g & J. Pharm. Belg., vol. 60, no. 6, 1978, page 397-399		

The present search report has been drawn up for all claims			TECHNICAL FIELDS SEARCHED (Int. Cl. 3)
			C 07 D 491/20 C 07 D 493/04 C 07 D 495/20
Place of search	Date of completion of the search	Examiner	
BERLIN	18-08-1983	PHILLIPS N.G.A.	
CATEGORY OF CITED DOCUMENTS			
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